

# **The International Philippines Natural Analogue Project (IPNAP)**

**- NA studies for bentonite reaction under hyperalkaline conditions -**

**The 12<sup>th</sup> NAWG Workshop**

**11<sup>th</sup>- 13<sup>th</sup> May, 2011, Cyprus**

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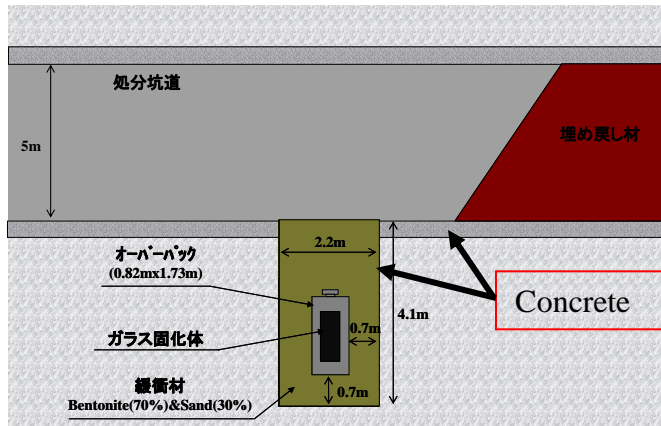
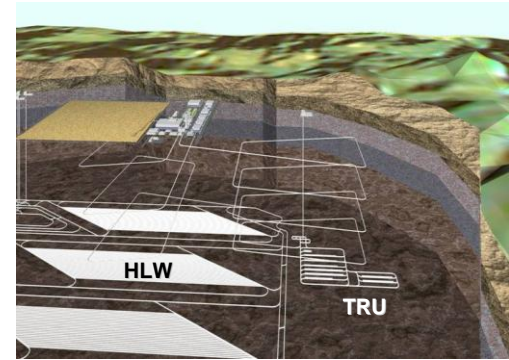
**R. Alexander(Bedrock Geosciences, Switzerland)**

# Contents

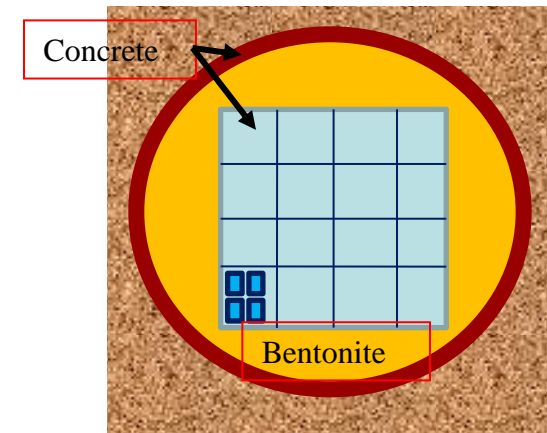
- 1. Background and Objectives**
- 2. Concept of Natural Analogue**
- 3. Geology and Groundwater Chemistry in Zambales area**
- 4. Field Evidences for Existence and Generation of the past hyperalkaline groundwater in Saile Deposits**
- 5. Formation of bentonite and Mineralogical characterization of bentonite**
- 6. Interaction between hyperalkaline groundwater and Smectite**
- 7. Alteration Process of Bentonite**
- 8. Summary and Future plan**

# 1. Background

- The long-term alteration phenomenon of bentonite using cement materials is a key issue of the waste disposal in Japan.



“HLW”



“TRU”

- Bentonite is unstable at high pH caused by cement leachates.

↳ Ophiolites have been identified as sources of high pH groundwaters.

- The reaction rates of between bentonite and high pH fluids are so slow as to be difficult to observe.

↳ Natural Analogues can be applicable in geological time scale.

# 1. Objectives

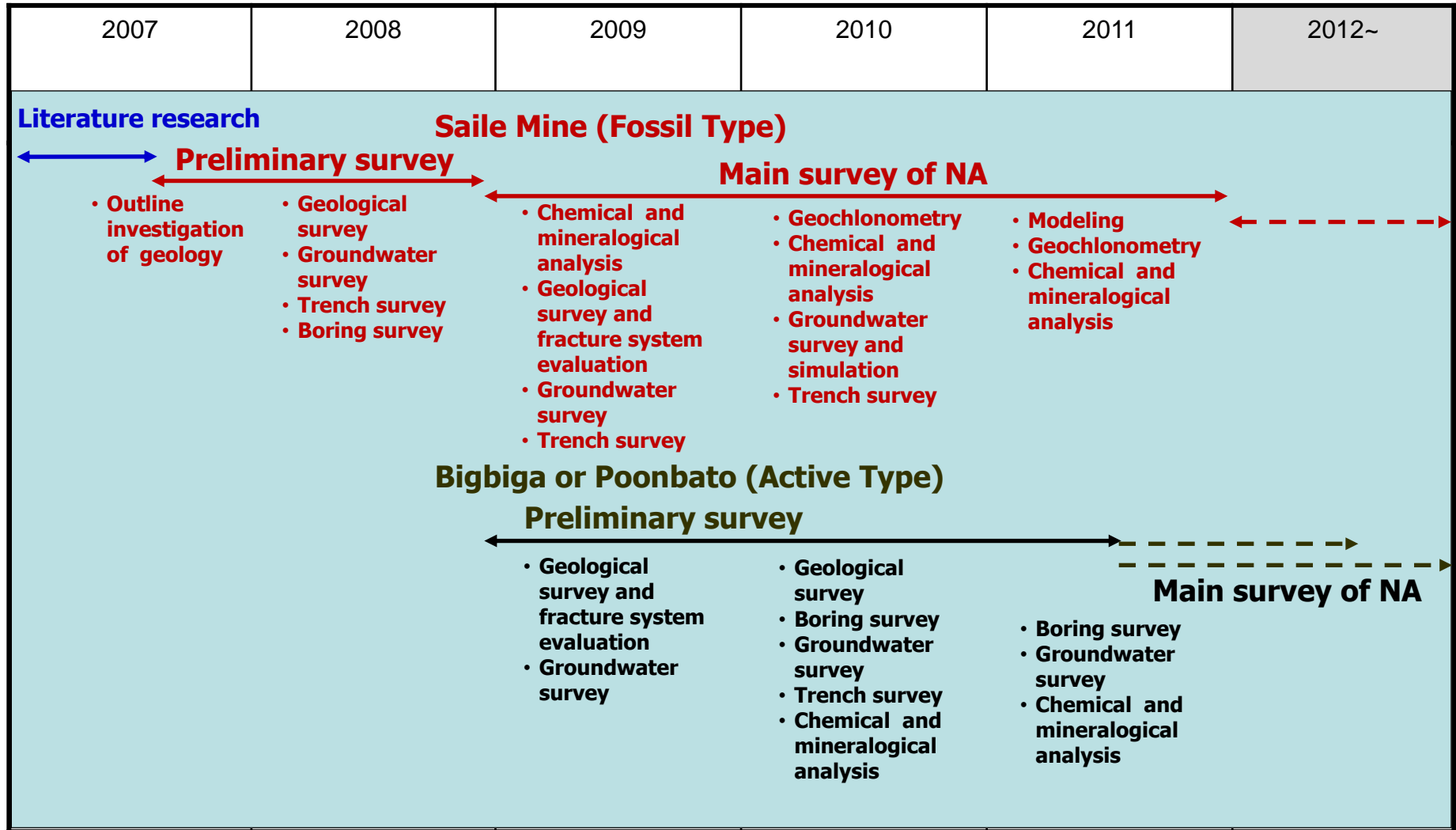
## Natural Analogues

- **NA study is to find out Natural Analogous phenomenon to understand Geochemical behavior of Bentonite Reaction, and evaluate long-term stability(longevity) of Bentonite under hyperalkaline conditions.**
- **NA could play a valuable role to bridge Disparity in Realism and Timescales between Laboratory Experiments and Systems represented in Repository Performance Assessment.**
- **NA could contribute Modelling to make a Robust Safety Case.**

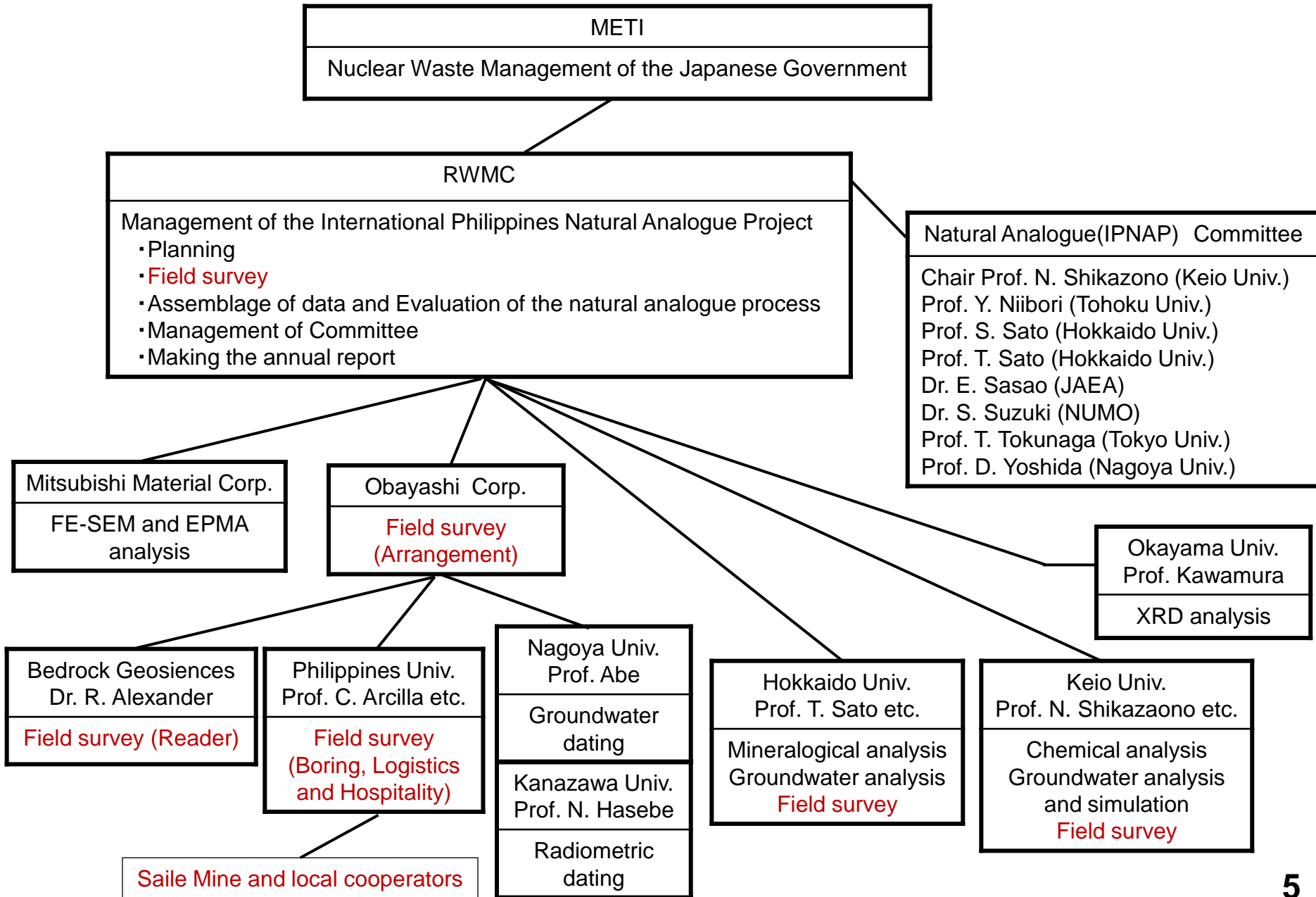
## Main task

- **To better understand Long-term stability of Bentonite in contact with Water Analogous to low alkali cement leachates**
  - **Key Process**
  - **Reaction Pathway**
  - **Environment Condition**
- **To establish comprehensive scenario of Bentonite Reaction**
- **To provide Data-sets for Improvement of Performance Assessment Modelling**

# 1. 1<sup>st</sup> phase IPNAP Plan (2007~2011)



# 1. Organization-System



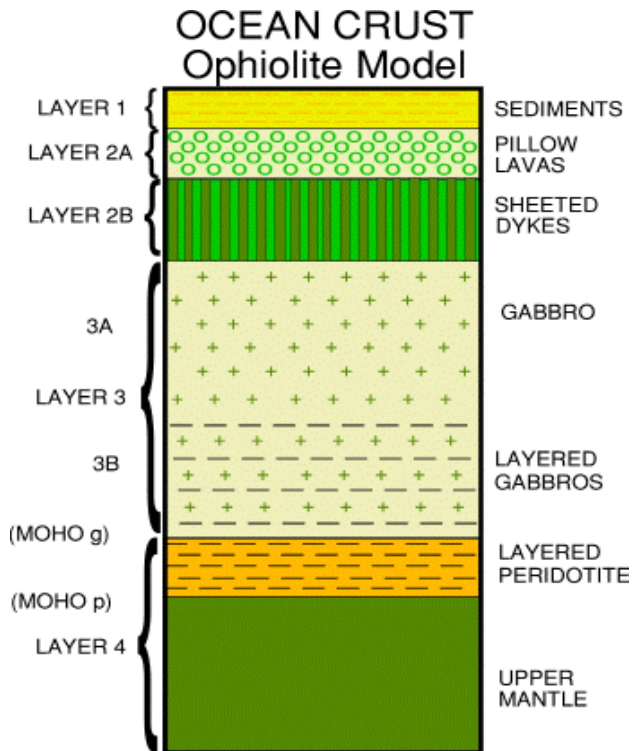
# 2. Concept of Natural Analogue

We focus on serpentinizing “Ophiolite” now as environment which generates hyperalkaline groundwater.

## Serpentinization of Ophiolite (Ultramafic Rocks)

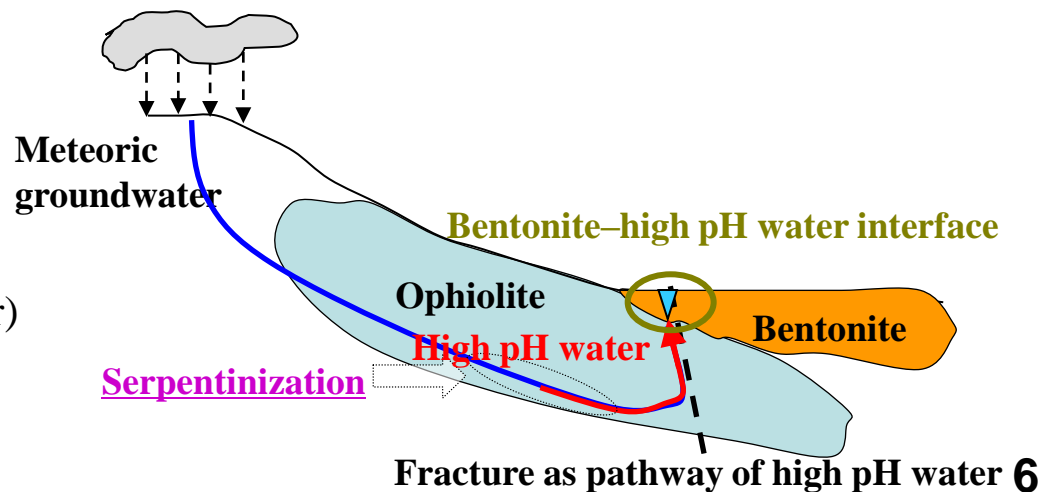
### Water-Ultramafic Rocks Interaction

- (1) Dissolution of Olivin / Pyroxenes
- (2) Precipitation of Serpentine Mineral Assemblages
- (3) Precipitation of Carbonates
- (4) Production of OH<sup>-</sup> (High pH condition)
- (5) Generation of Gas mixtures
- (6) Exothermic Reaction (Hot Spring)
- (7) Reducing Condition

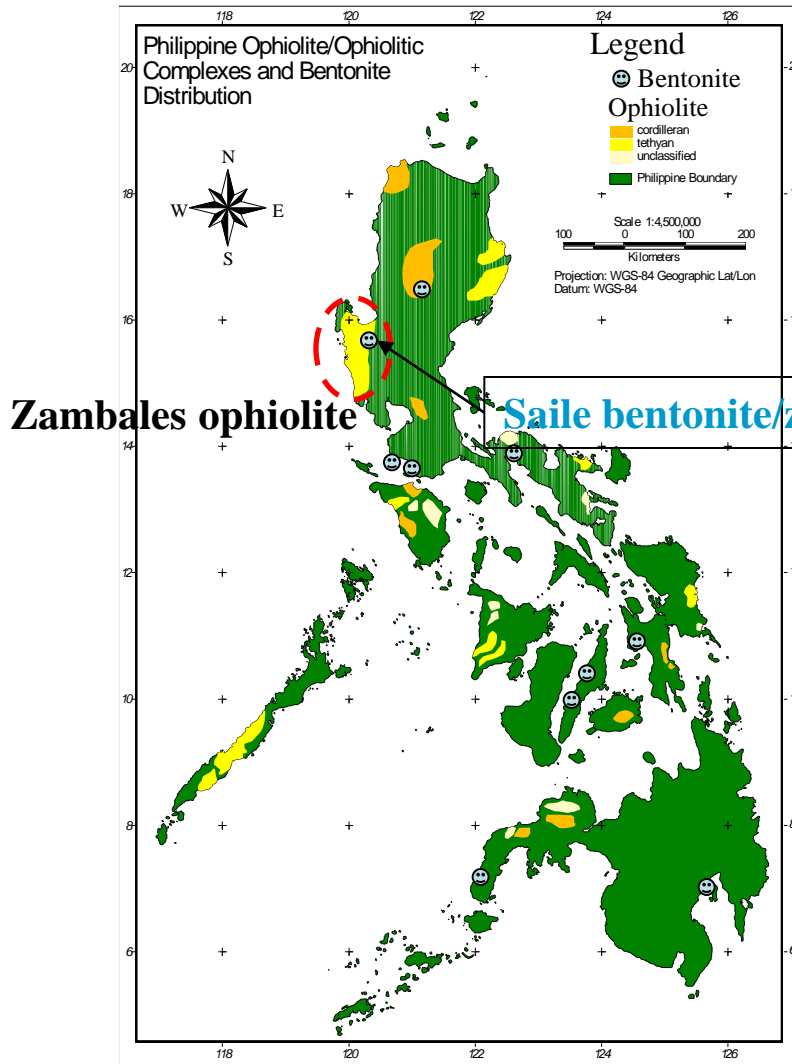


### Essential components for supporting concept of NA

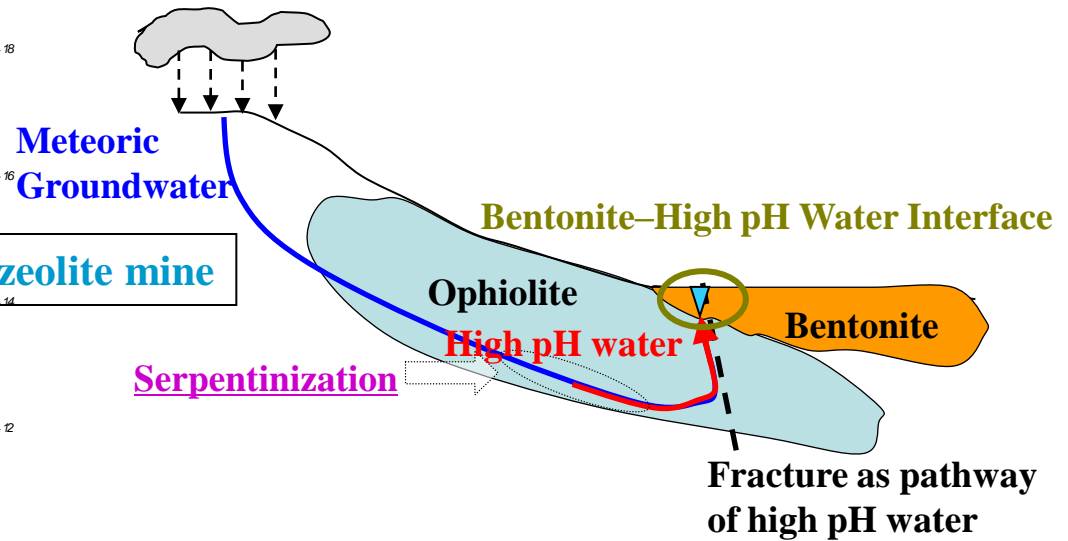
- (1) Bentonite layer (Bed)
- (2) High pH water (Fluid)  
(Generation of hyperalkaline groundwater)
- (3) Pathway of High pH water  
(Bentonite – high pH water interface)



## 2. Why Philippines ?



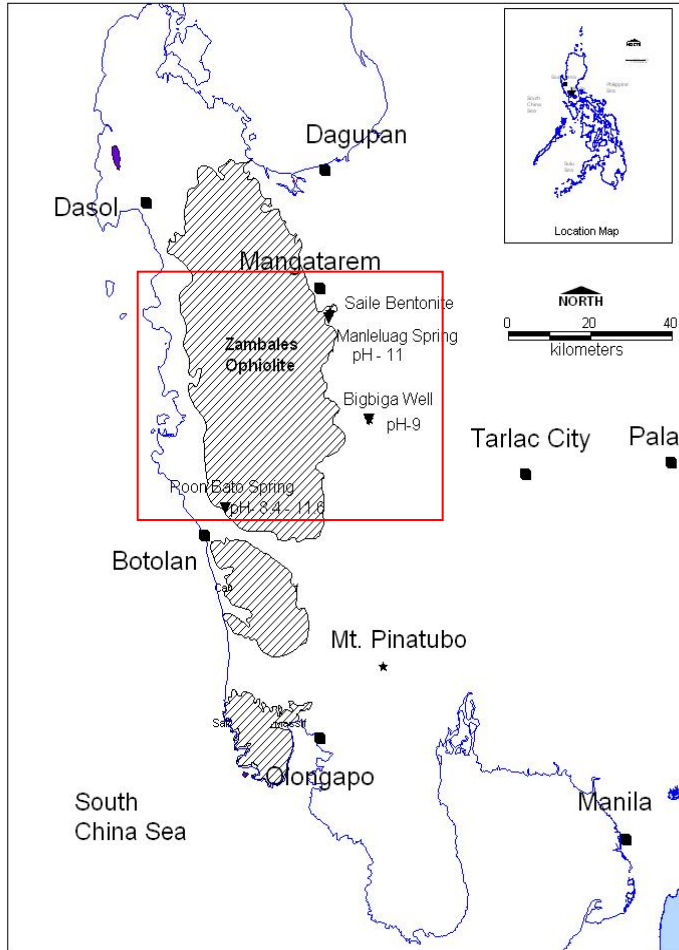
### The Natural Analogue Concept in the Philippines



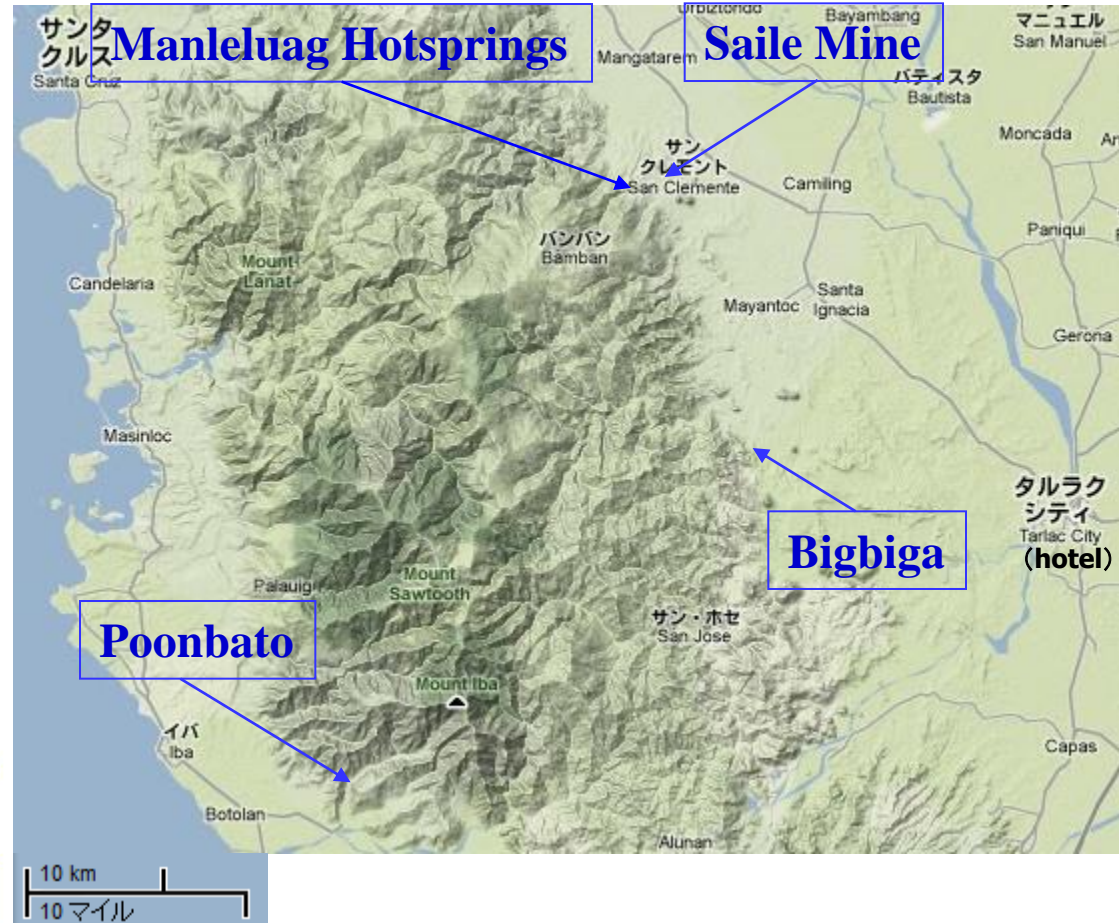
The Philippines was selected as research site because of existence of bentonite deposits near hyperalkaline springs in the ophiolite basement.

Distribution of Ophiolite and Bentonite deposits in the Philippines

# 3. Survey Area



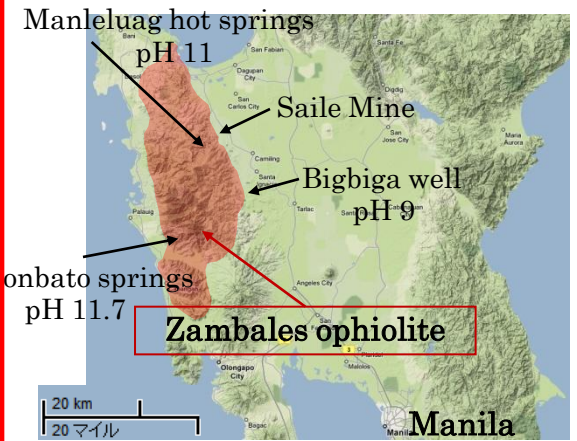
**Zambales Ophiolite**



**Survey site**  
**(West central area of the island of Luzon)**

# 3. Groundwater Chemistries of Survey Site

| Site                                | Manleluag HSp. - M1 | Manleluag HSp. - M2 | Manleluag HSp. - M3 | Saile Mine   | Poonbato     | Bigbiga Well | low alkali cement leachates (PNC 1997)* |
|-------------------------------------|---------------------|---------------------|---------------------|--------------|--------------|--------------|---|
| Sample No.                          | MA-10-HWP-01        | MA-10-HWP-02        | M3-10-HWP-03        | SM-10-HWP-01 | PB-10-HWP-01 | BB-10-HWP-06 |   |
| pH                                  | 11.08               | 10.75               | 10.83               | 6.73         | 11.69        | 9.09         | 11.09                                   |
| ORP(Eh) [mV]                        | -713                | -682                | -496                | 65           | 56           | 252          | -                                       |
| Temp [°C]                           | 34                  | 34.1                | 34                  | 27.9         | 27.4         | 30.1         | 60                                      |
| CH <sub>4</sub> [ppm]               | 1140                | 470                 | 2200                | 0            | >5000        | 0 (0~560)    | -                                       |
| H <sub>2</sub> [ppm]                | 150                 | 53                  | 230                 | 0            | 50~1320      | 0 (0~130)    | -                                       |
| Na <sup>+</sup> [ppm]               | 24.0                | 24.9                | 25.1                | 11.4         | 21.6         | 107          | 43                                      |
| K <sup>+</sup> [ppm]                | 5.24                | 0.764               | 0.045               | nd           | 36.4         | 0.215        | 13                                      |
| Ca <sup>2+</sup> [ppm]              | 323                 | 68.3                | 95.4                | 84.4         | 182          | 0.599        | 16.8                                    |
| Mg <sup>2+</sup> [ppm]              | 2.98                | 0.054               | 0.06                | 44.1         | 0.147        | 0.092        | -                                       |
| Al [ppm]                            | 0.463               | 0.242               | 0.336               | 0.02         | 0.103        | 0.012        | 0.3                                     |
| Cl <sup>-</sup> [ppm]               | 6.98                | 6.88                | 7.14                | 0.703        | 0.226        | 1.543        | -                                       |
| SO <sub>4</sub> <sup>2-</sup> [ppm] | 0.502               | 0.413               | 0.330               | 0.081        | 0            | 1.332        | -                                       |



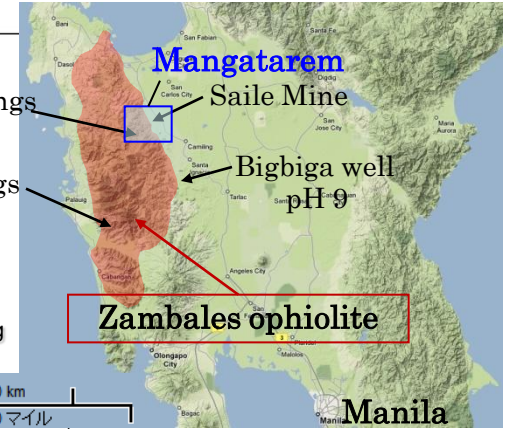
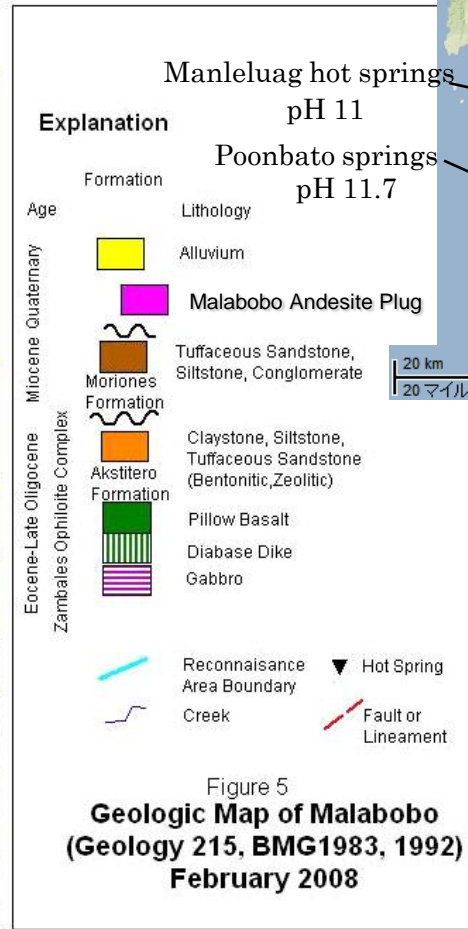
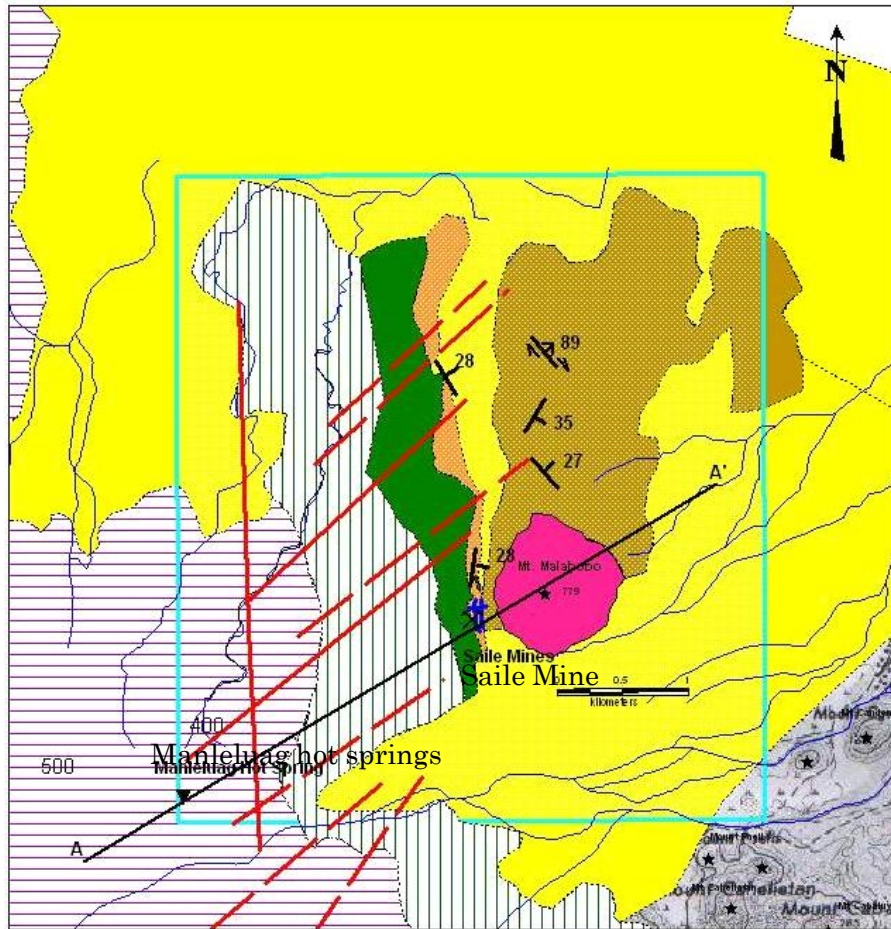
Manleluag Hsp.- M1    Manleluag Hsp.- M3



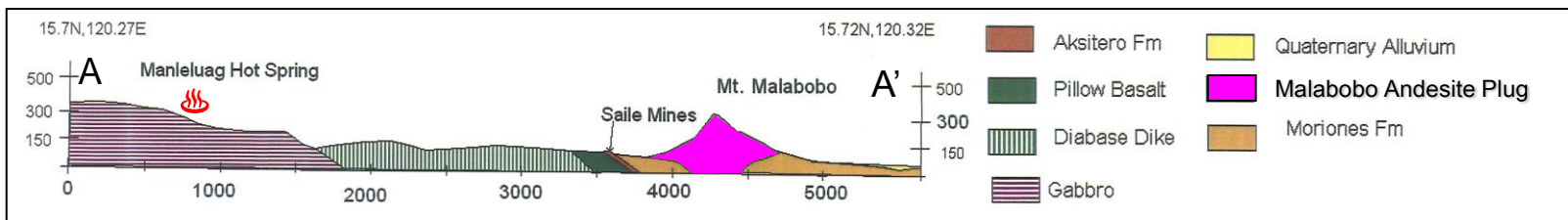
The groundwater chemistry is analogous to low-alkali cement leachates.

\*Mix Proportion – Portland cement: silica fume: fly ash = 40: 20: 40

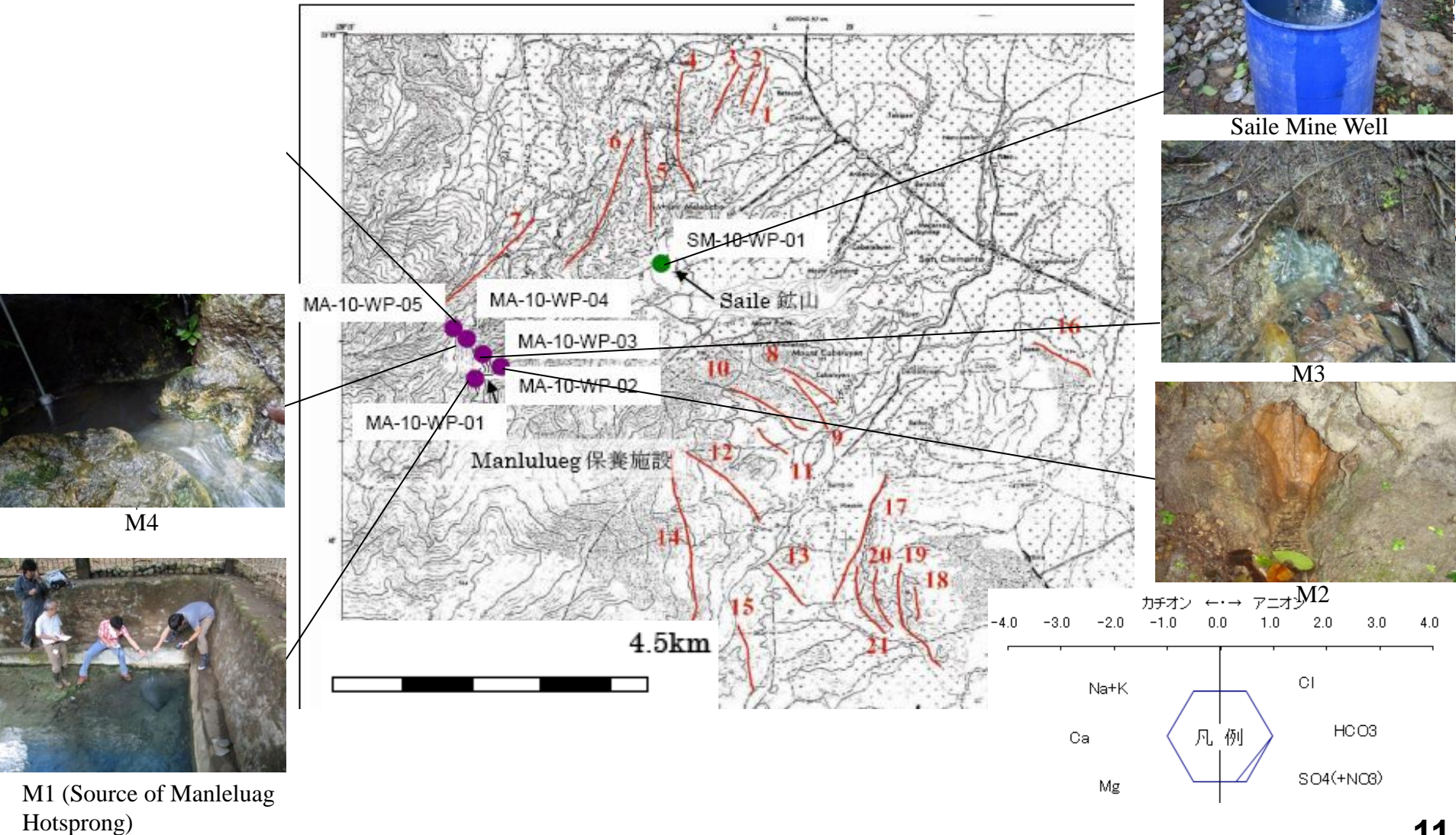
# 3. Geological map of Mangatarem (Manleluag Hotspring / Saile Mine)



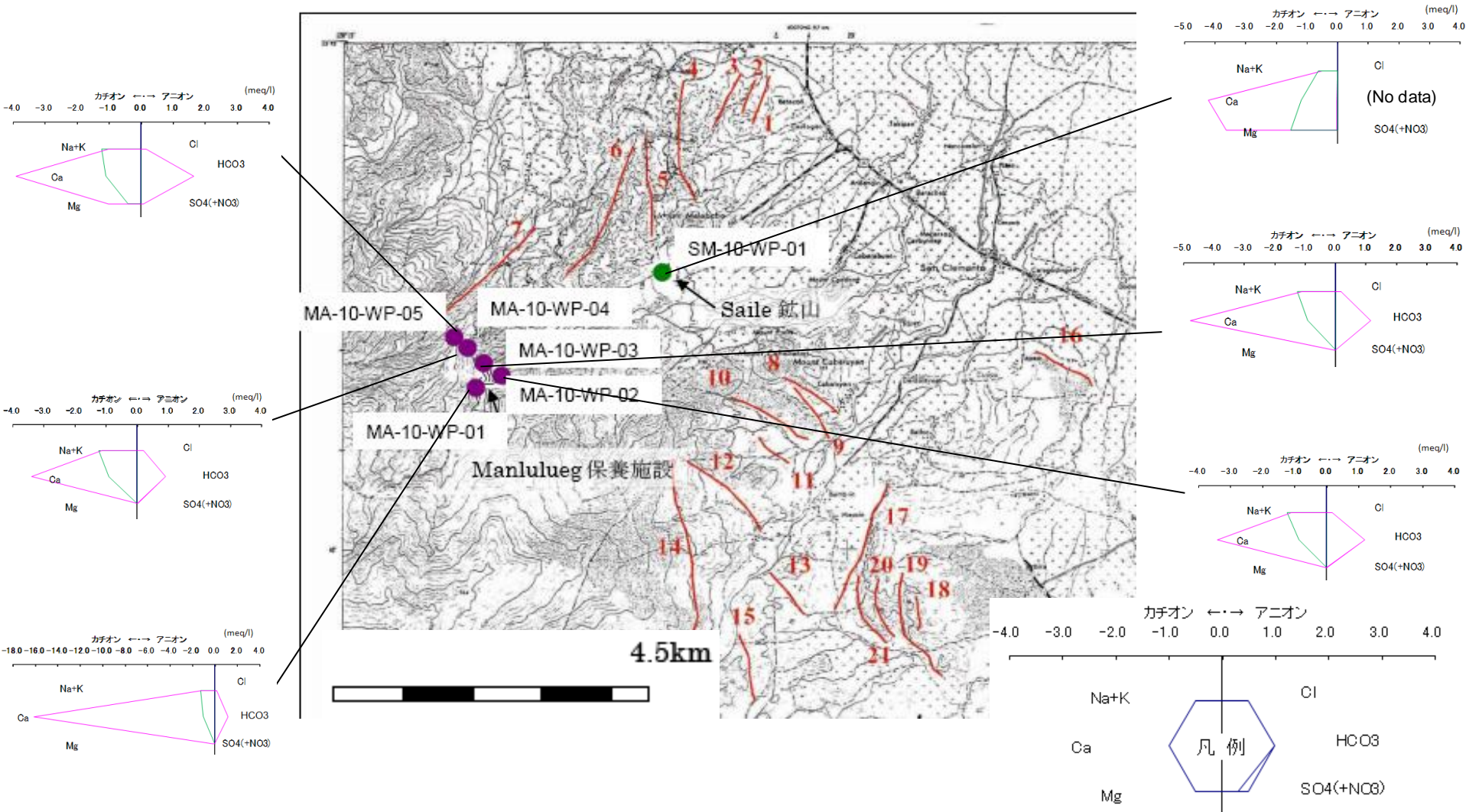
Saile Mine Quarry



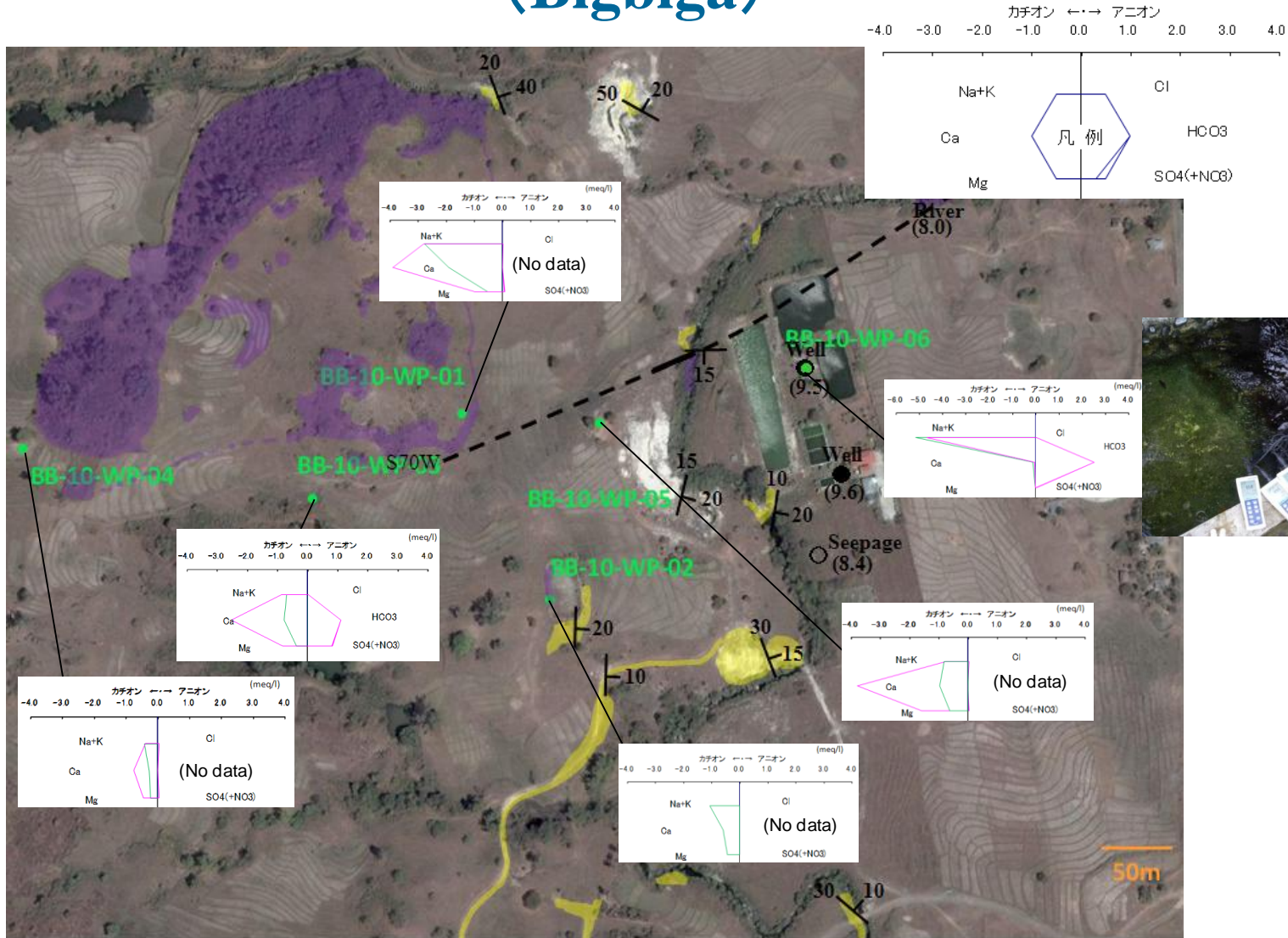
# 3. Groundwater Chemistries (Manleluag Hotspring / Saile Mine)



# 3. Groundwater Chemistries (Manleluag Hotspring / Saile Mine)



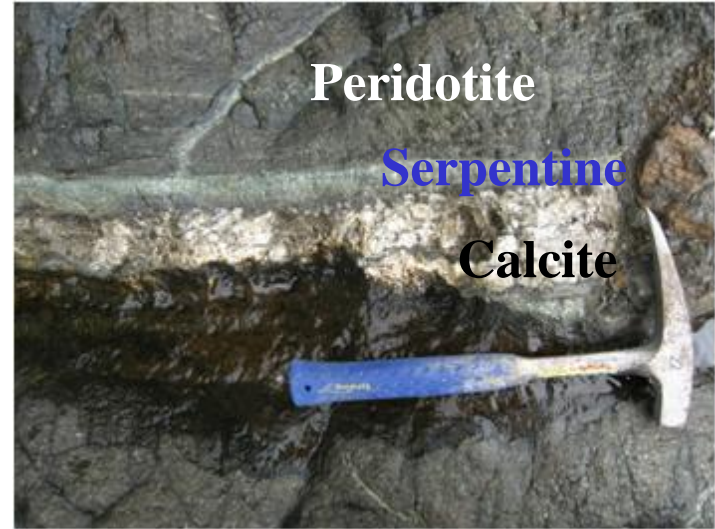
# 3. Groundwater Chemistries (Bigbiga)



### 3. Outcrop (Poonbato)



**Hyperalkaline groundwater discharge point  
(pH11.6)**



**Serpentinaization**

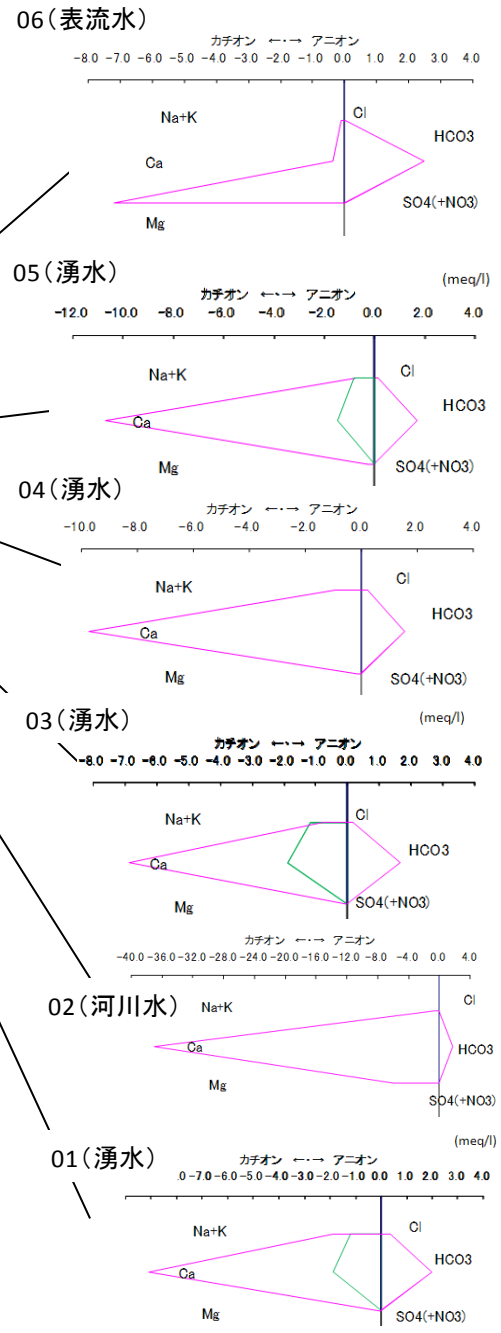
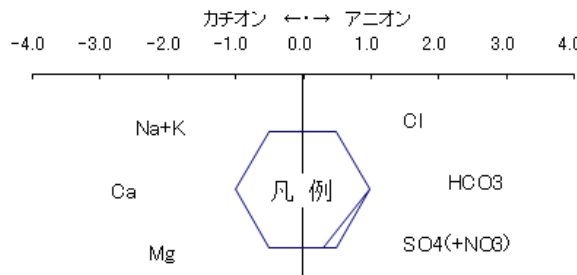
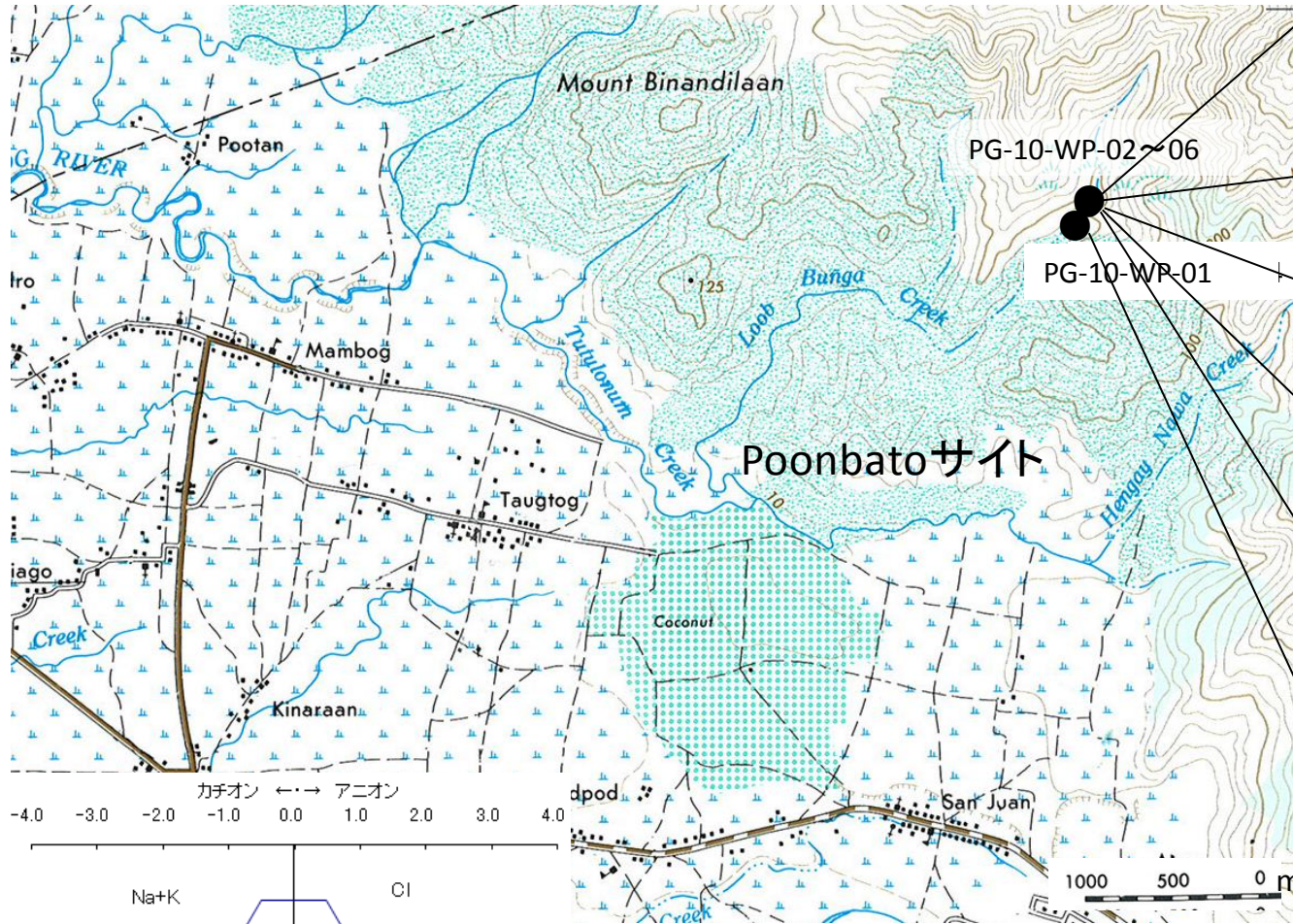


**Travertine**



**Natural concrete**

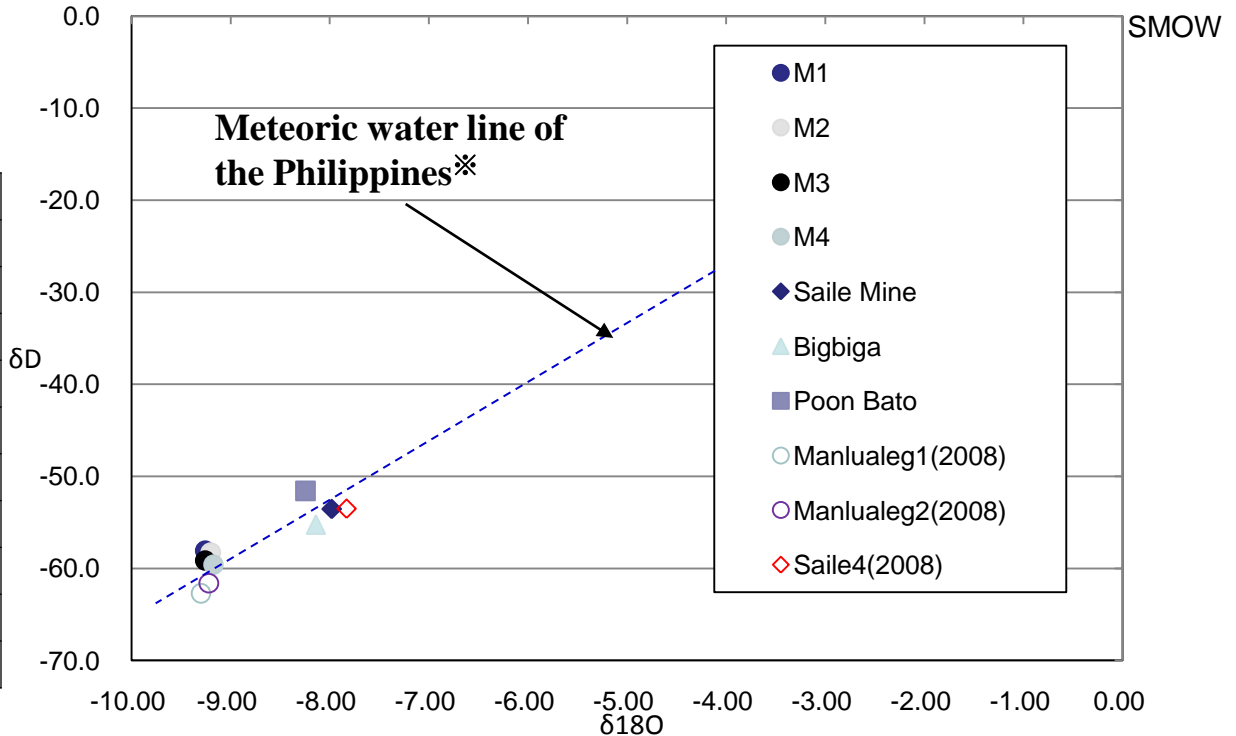
# 3. Groundwater Chemistries (Poonbato)



# 3. Origin and Residence Time of Groundwater

## Stable Isotope Ratio of Rain Water and Groundwater

|                  |    | $\delta D$ | $\delta^{18}O$ |
|------------------|----|------------|----------------|
| Manleluag        | M1 | -58.1      | -9.26          |
|                  | M2 | -58.3      | -9.20          |
|                  | M3 | -59.1      | -9.26          |
|                  | M4 | -59.6      | -9.17          |
| Saile Mine       |    | -53.5      | -7.98          |
| Bigbiga          |    | -55.2      | -8.14          |
| Poonbato         |    | -51.6      | -8.24          |
| Manlualeg1(2008) |    | -62.7      | -9.30          |
| Manlualeg2(2008) |    | -61.6      | -9.22          |
| Saile4(2008)     |    | -53.5      | -7.83          |



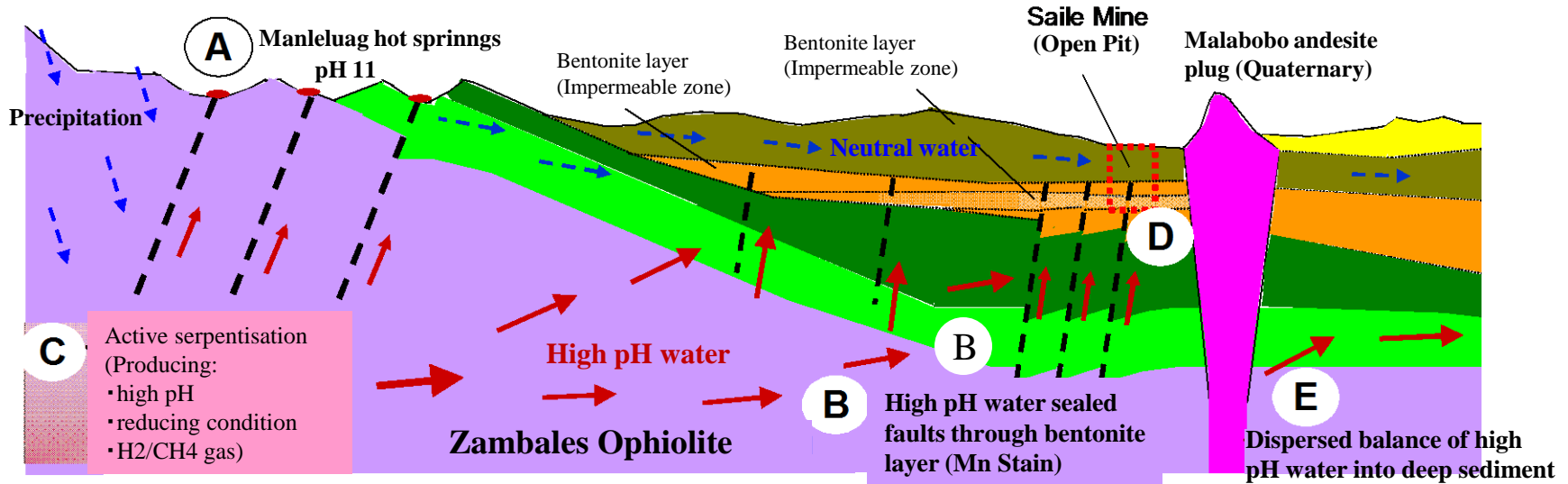
※From Abaya (2005)

- Meteoric water origin
- Around 1000m of Recharge Elevation

## Groundwater Dating by $^{14}C$ method

- Quantification by Tandem Accelerator Mass Spectrometry in Tono
- About 3000 years of residence time of groundwater at Bigbiga even if the oldest

# 3. Basic Concept of Regional Groundwater Flow System in Mangatarem



## Aksitero Formation

- ▼ Occurrence (Width: 30~150m, Strike: about 2000m around Saile mine)
- ▼ Lithofacies
  - Medium-grained sandstone, siltstone, claystone accompanied by tuffaceous (volcanic ashes: pumice, volcanic glass) lamina
  - Tuffaceous lamina and limestone accompanied by tuffaceous fine-grained turbidite layer
  - Calcareous and tuffaceous sandstone
- ▼ Mineral Composition of Bentonite/Zeolite deposits (Saile Mine)
  - Main mineral: Ca-Montmorillonite
  - accessory mineral: Serpentine
    - Zeolite (Heulandite, Clinoptilolite, Mordenite)
    - Calcite
    - Manganite
    - Magnetite
    - Quartz,
    - Plagioclase
    - Xenocrysts of Chromite, Clinopyroxene

| General Stratigraphy |   |
|----------------------|---|
| Quaternary           | Alluvium  |
|                      | Andesite plug   |
| ~~~~~ <unconformity> |   |
| Miocene              | <b>「Moriones Formation」</b>                                       |
|                      | Tuffaceous sandstone, Siltstone, Conglomerate                     |
| ~~~~~ <unconformity> |   |
| Late Oligocene       | <b>「Aksitero Formation」</b>                                       |
|                      | Claystone, Sandstone, Tuffaceous sandstone (Bentonitic, Zeolitic) |
| Eocene               | <b>「Zambales Ophiolite」</b>                                       |
|                      | Pillow lava   |
|                      | Diabase dike  |
|                      | Gabbro  |



# 4. Field Evidences for Existence and Generation of Hyperalkaline Groundwater in Saile Deposits

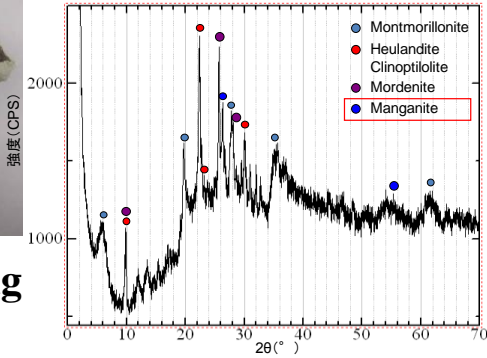
## (1) Mn-Staining (Mn-precipitation: Black)



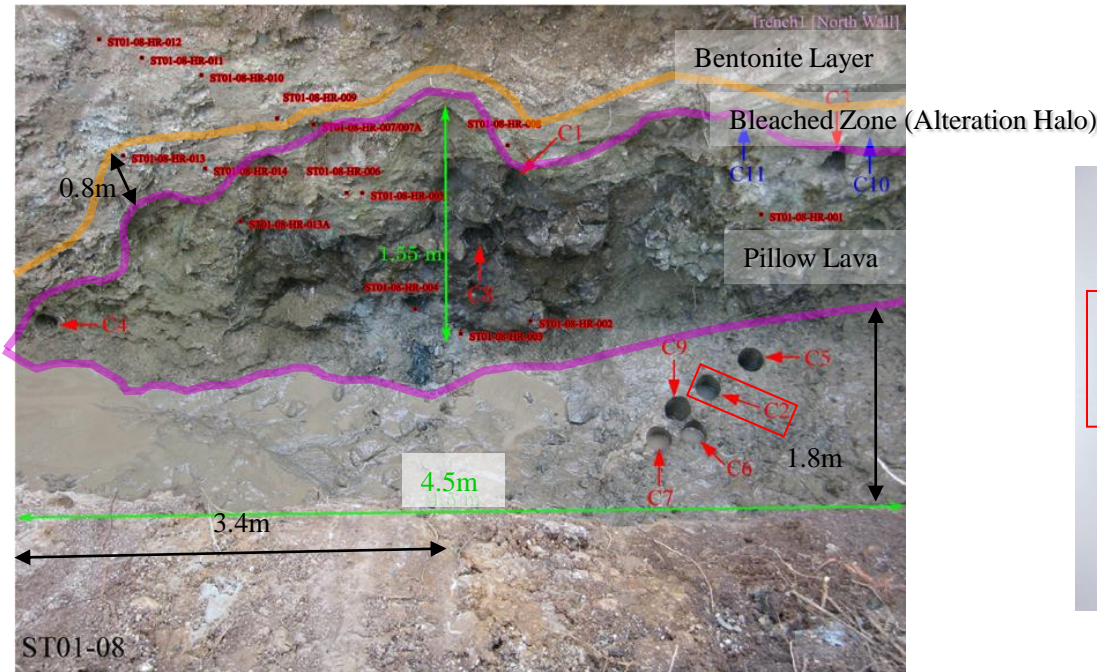
**【Quarry】**



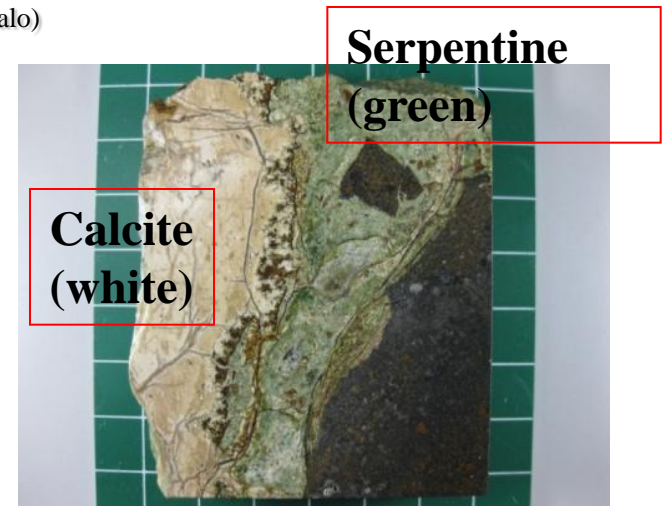
**Fracture filling materials (F<sub>3</sub>)**



**Manganite( $\gamma$ -MnOOH)**



**【Trench 1】**



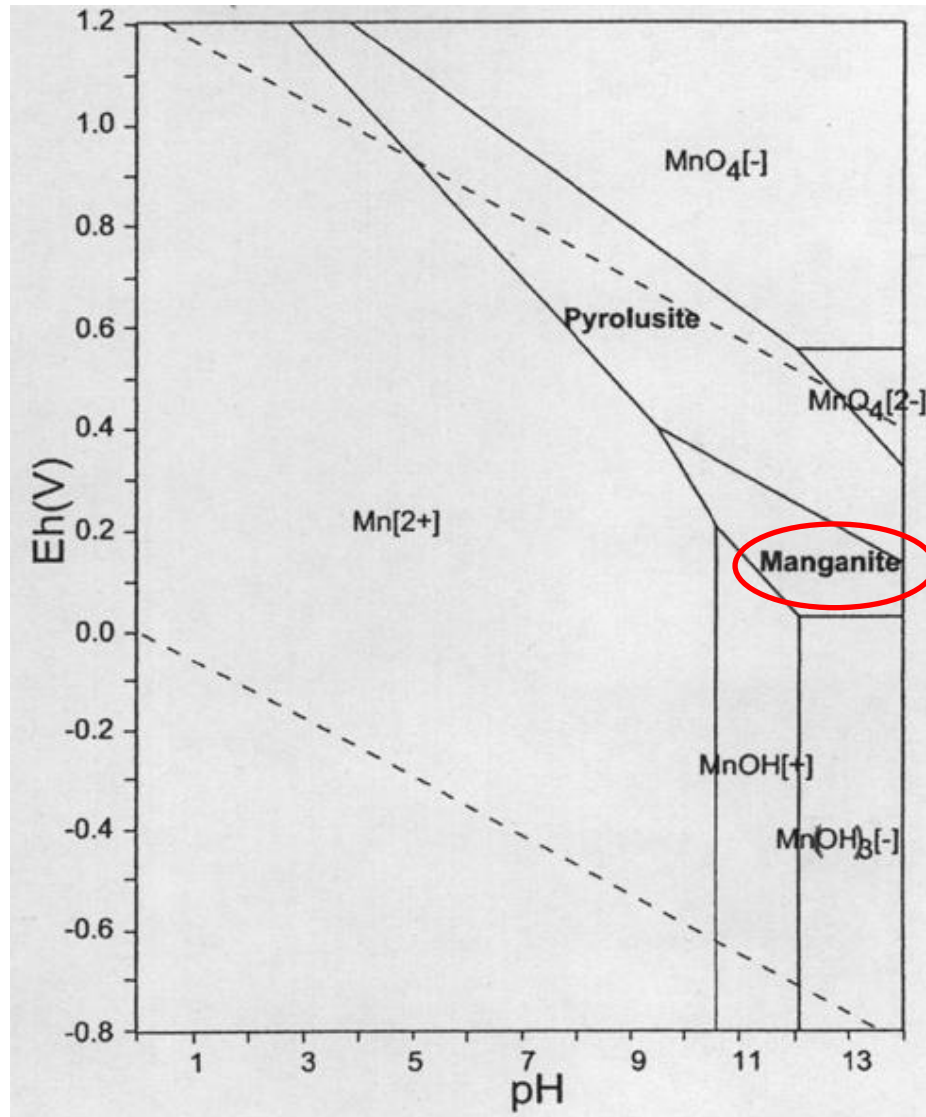
(Inner)

(Outer)

**(2) Fracture filling materials (C<sub>2</sub>)**

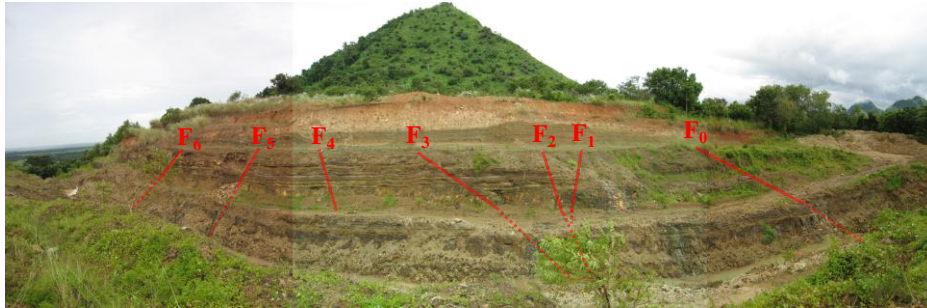
# 4. Phase Diagram of Mn-oxides and Mn-hydroxides

(JNC-TDB 011213g2 / GWB)

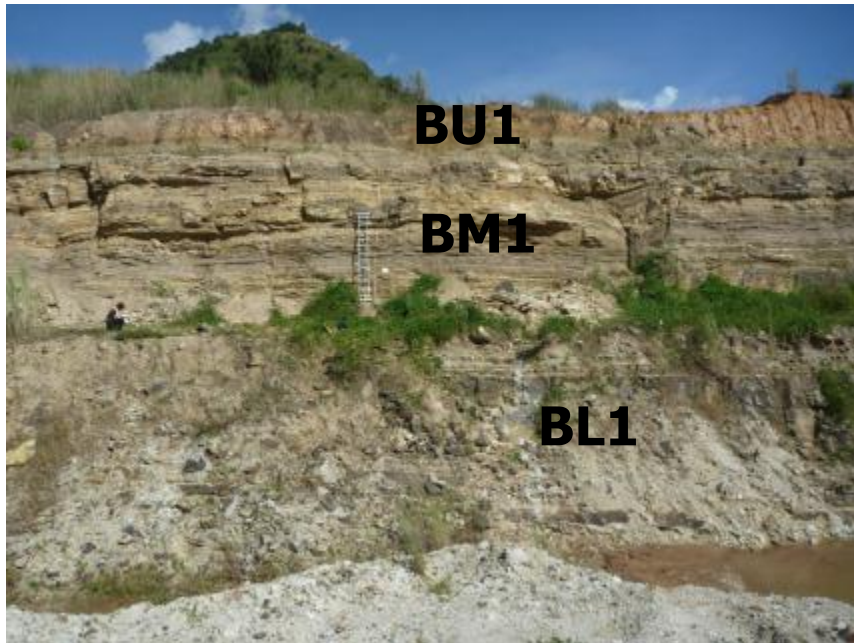


# 5. Formation of bentonite

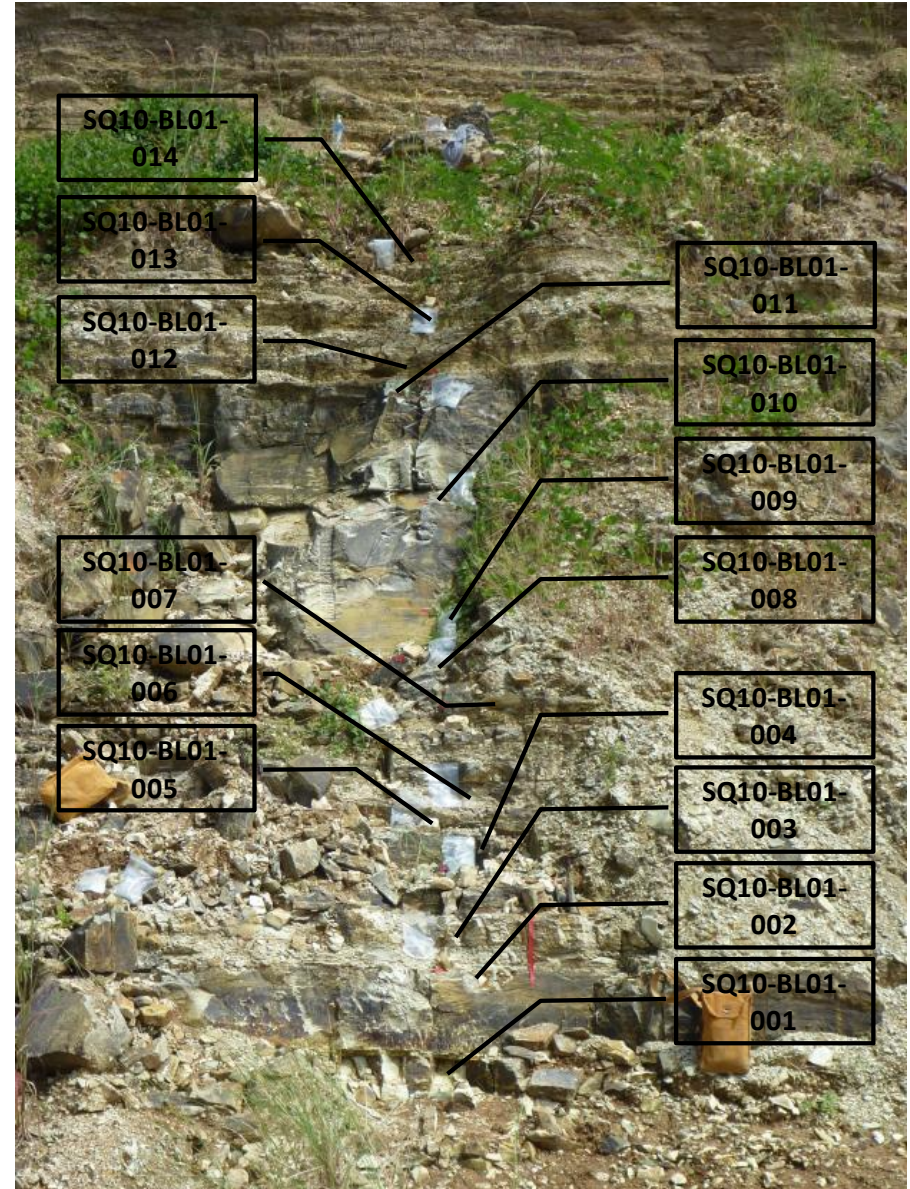
Systematic sampling of bentonite/zeolite layer – Baseline 1



Quarry



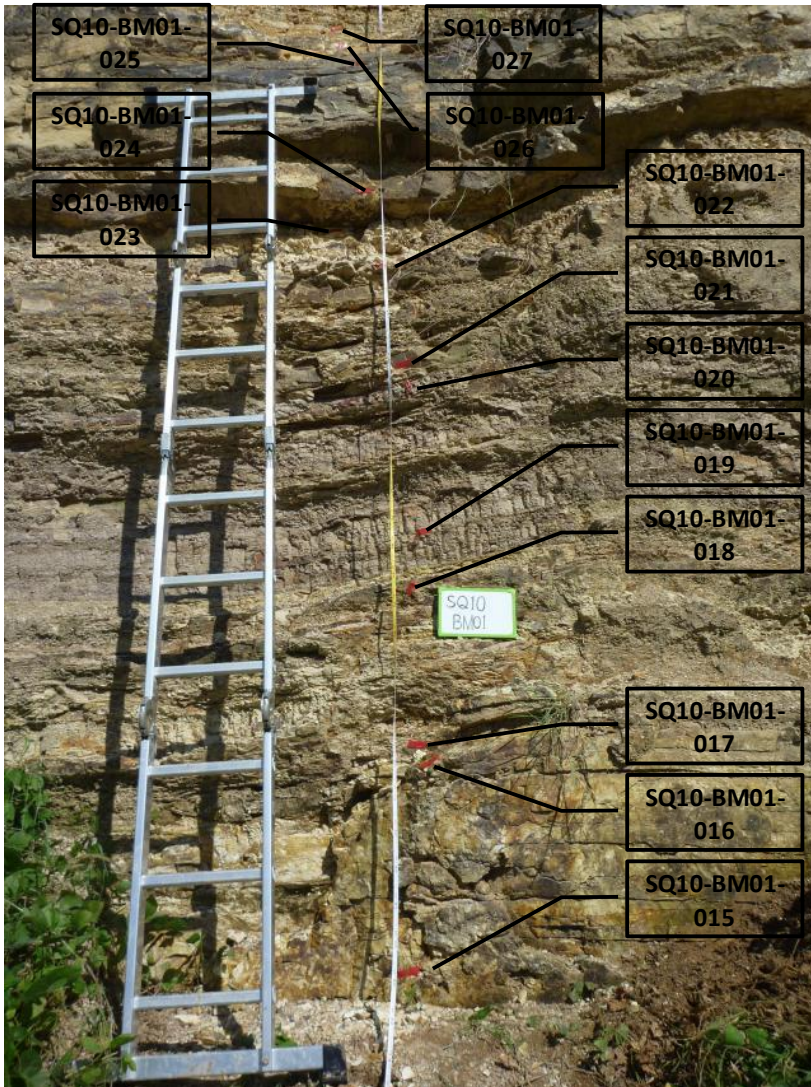
3 parts of Baseline 1



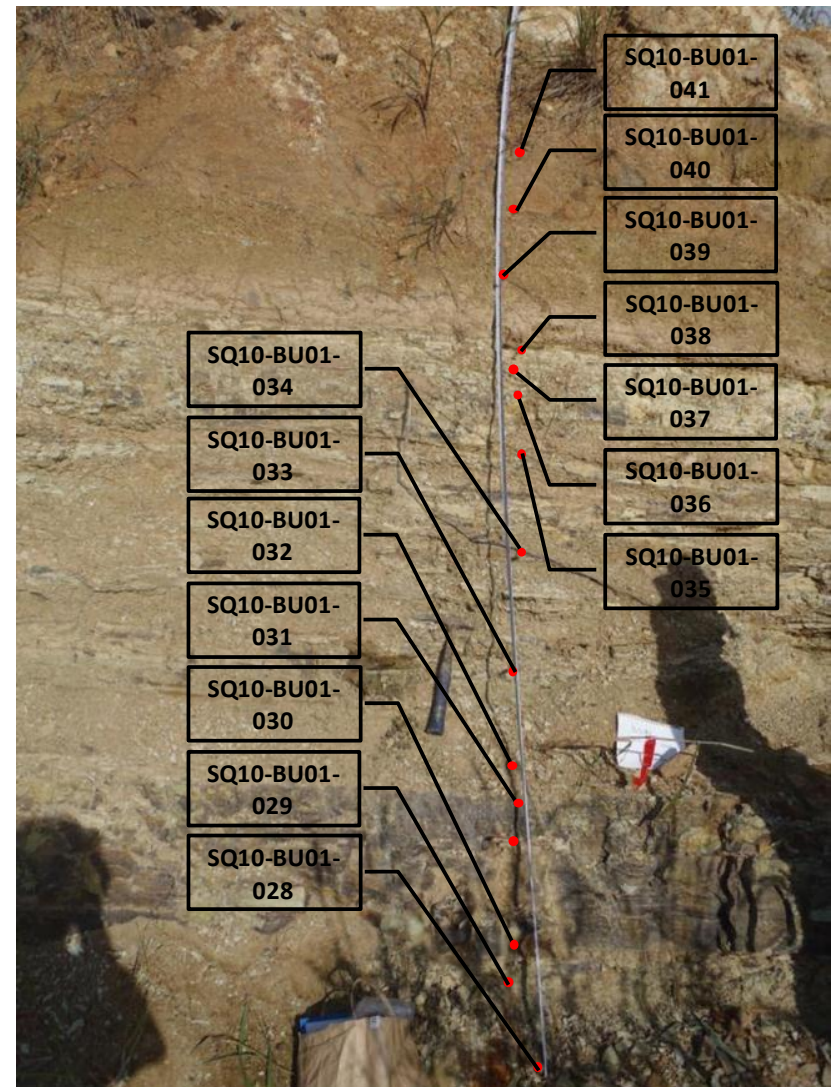
Sample location of BL1 (SQ-10-BL1-001~014)

# 5. Formation of bentonite

Systematic sampling of bentonite/zeolite layer – Baseline 1



Sample location of BM1  
(SQ-10-BM1-015~027)

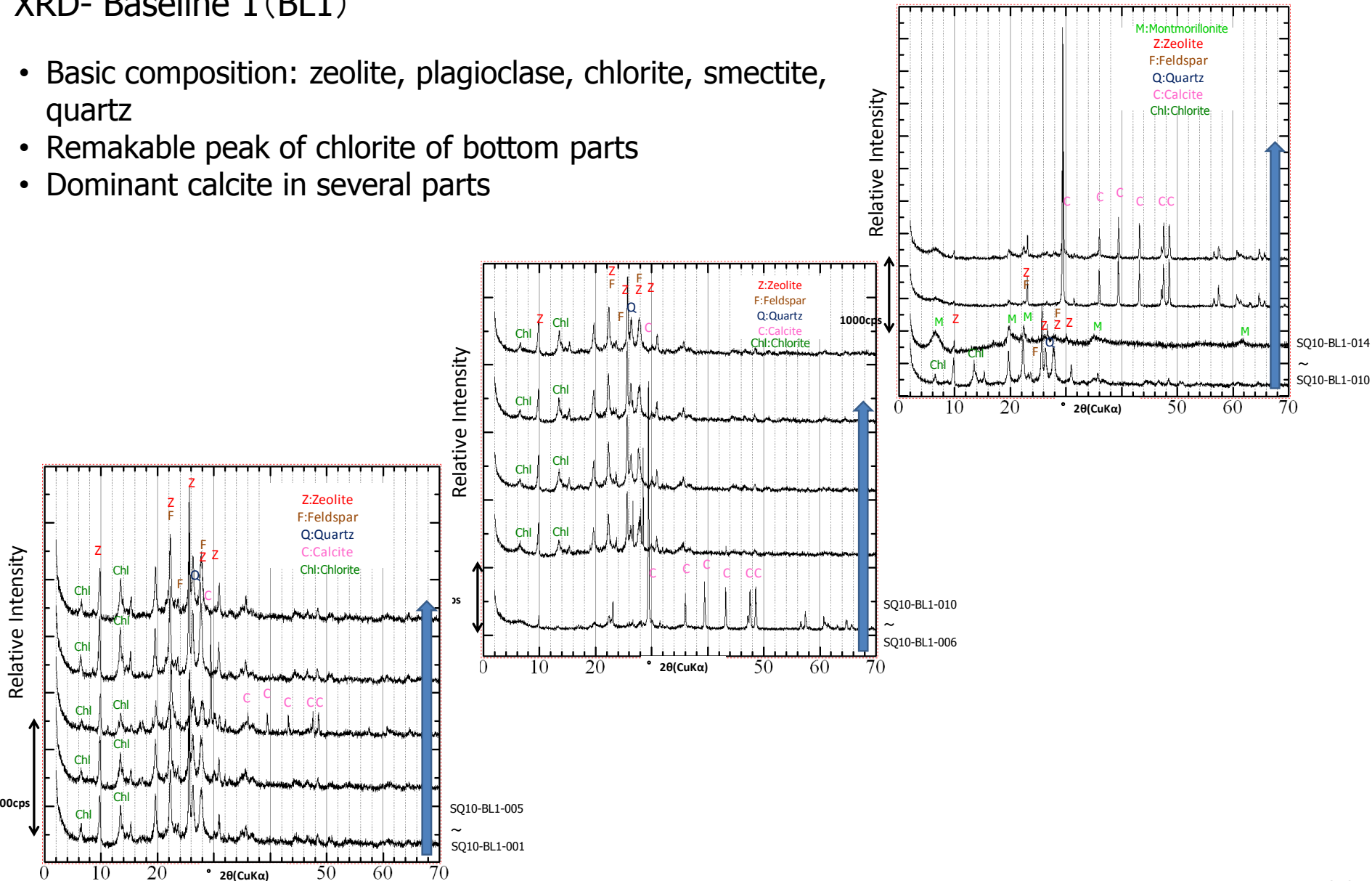


Sample location of BU1  
(SQ-10-BU1-028~041)

# 5. Formation of bentonite

## XRD- Baseline 1 (BL1)

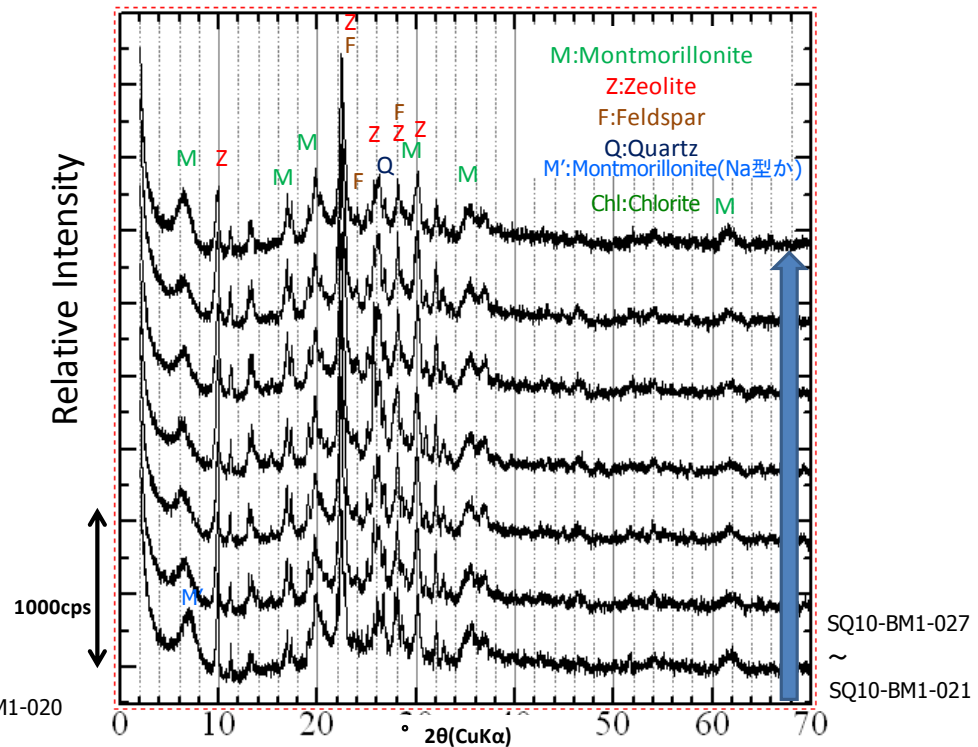
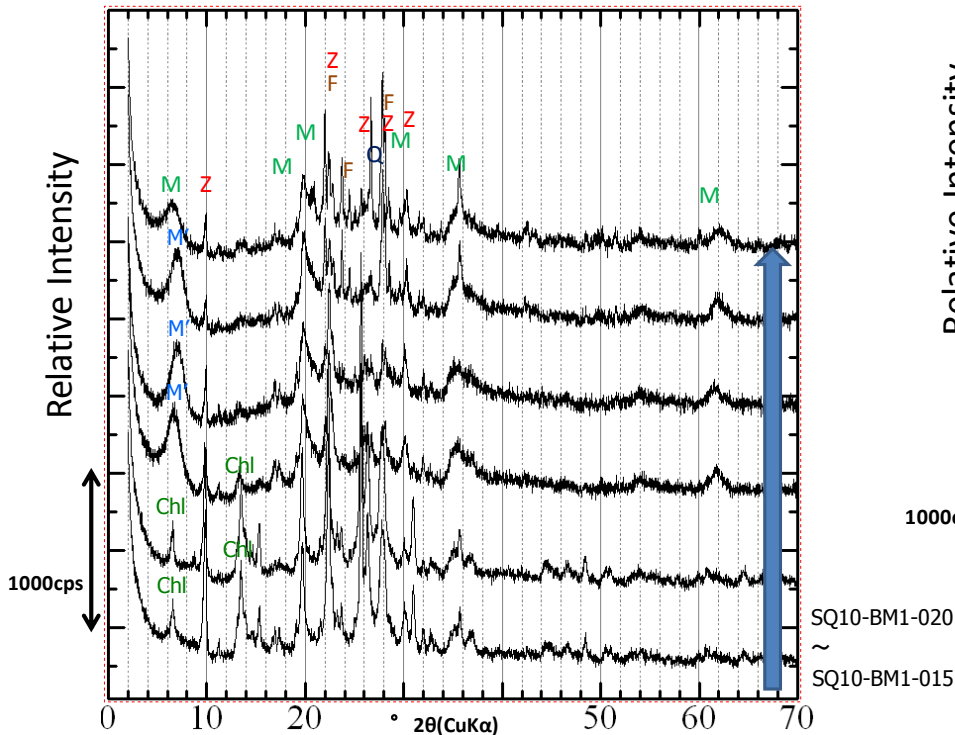
- Basic composition: zeolite, plagioclase, chlorite, smectite, quartz
- Remakable peak of chlorite of bottom parts
- Dominant calcite in several parts



# 5. Formation of bentonite

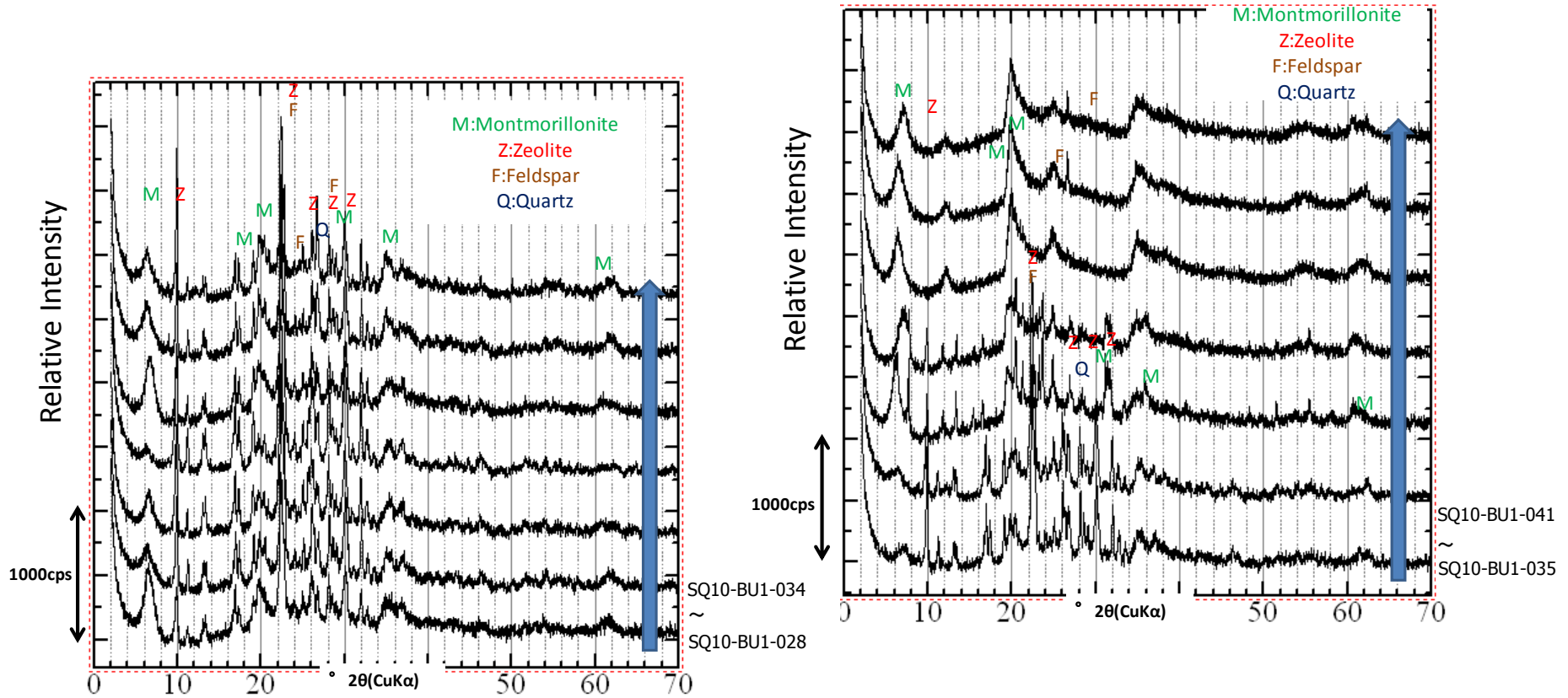
## XRD- Baseline 1 (BM1)

- Basic composition: zeolite, plagioclase, smectite (Disappearance of Chlorite), quartz
- Dominant zeolite in several parts
- Different peak of 001 with each horizon (Different interlayer cation of smectiite)  
→XRD under relative humidity control



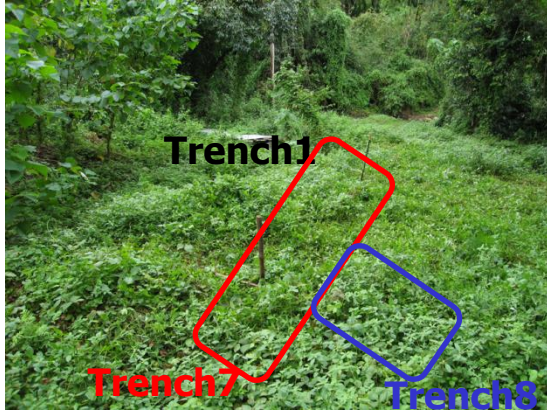
# 5. Formation of bentonite

XRD- Baseline 1 (BU1)



# 5. Trench Survey

## Trench location (Saile Mine)

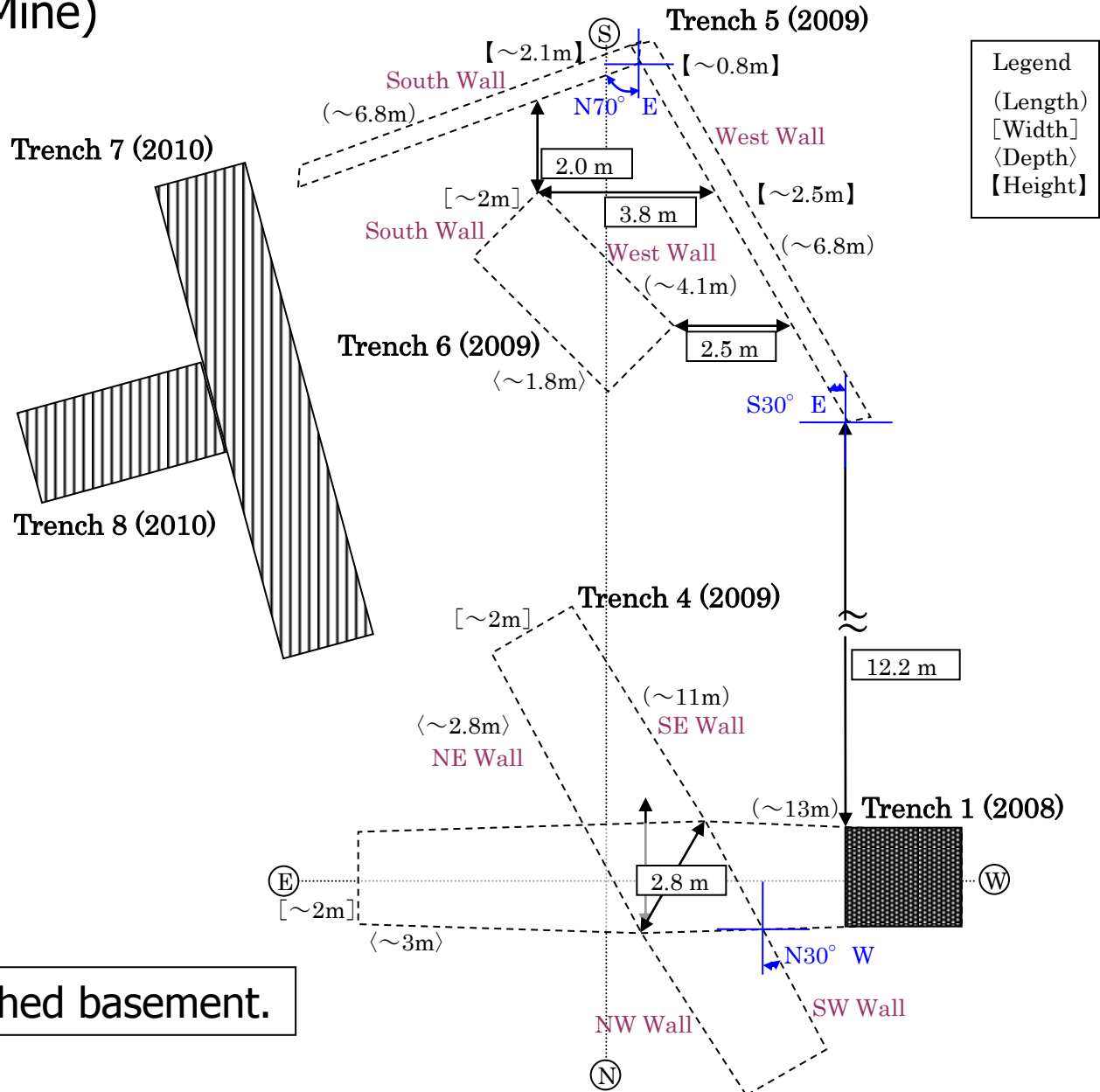


Before digging



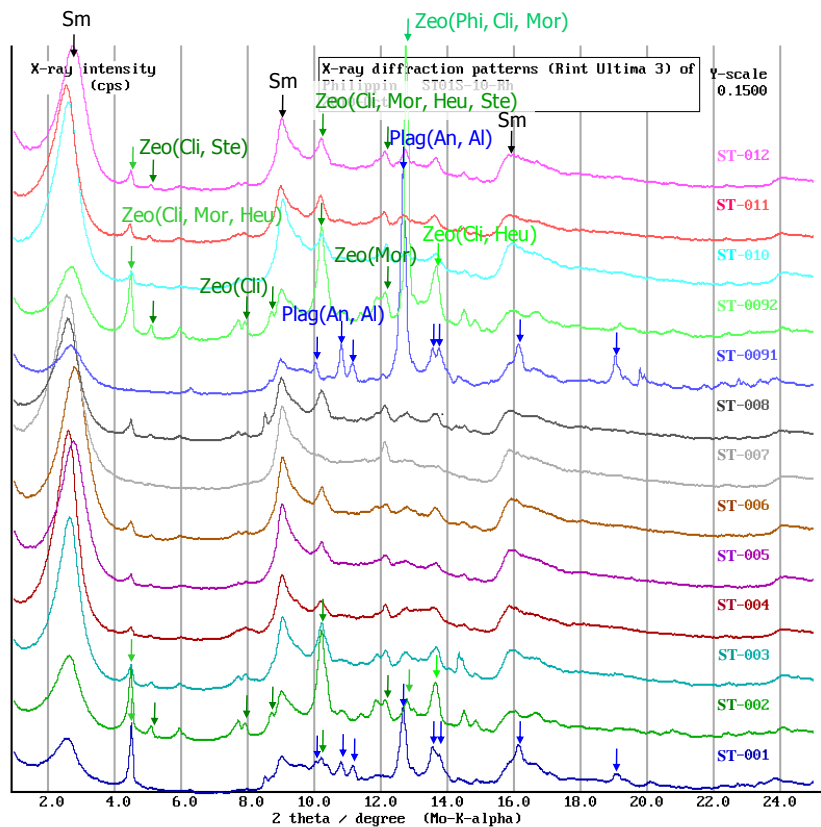
Digging of Trench 7

Trench 1, 5, 6 reached basement.



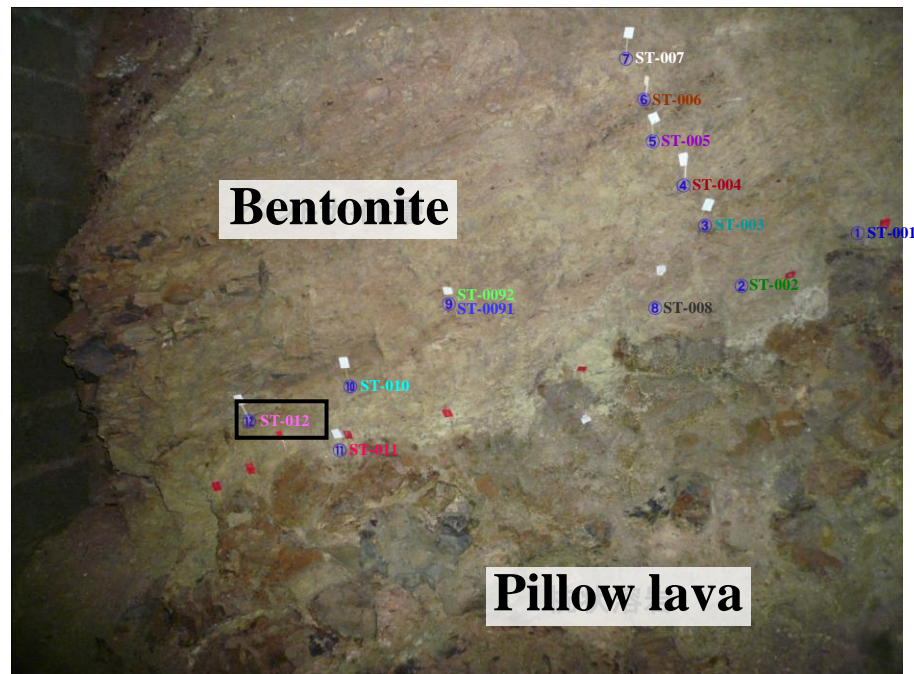
# 5. Mineralogical characterization of Bentonite

## 1. Mineral Composition of Trench 1 South Wall

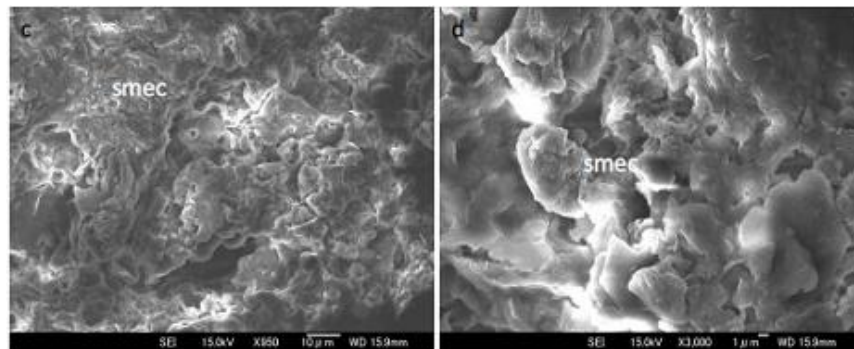


Sm:Smectite, Zeo:Zeolite (Cli: Clinoptilolite, Mor: Mordenite, Heu: Heulandite, Phi: Phillipsite, Ste: Stellerite), Qz:Quartz, Cal:Calcite, Plag:Plagioclase

### XRD Results (Tokyo Institute of Tech.)



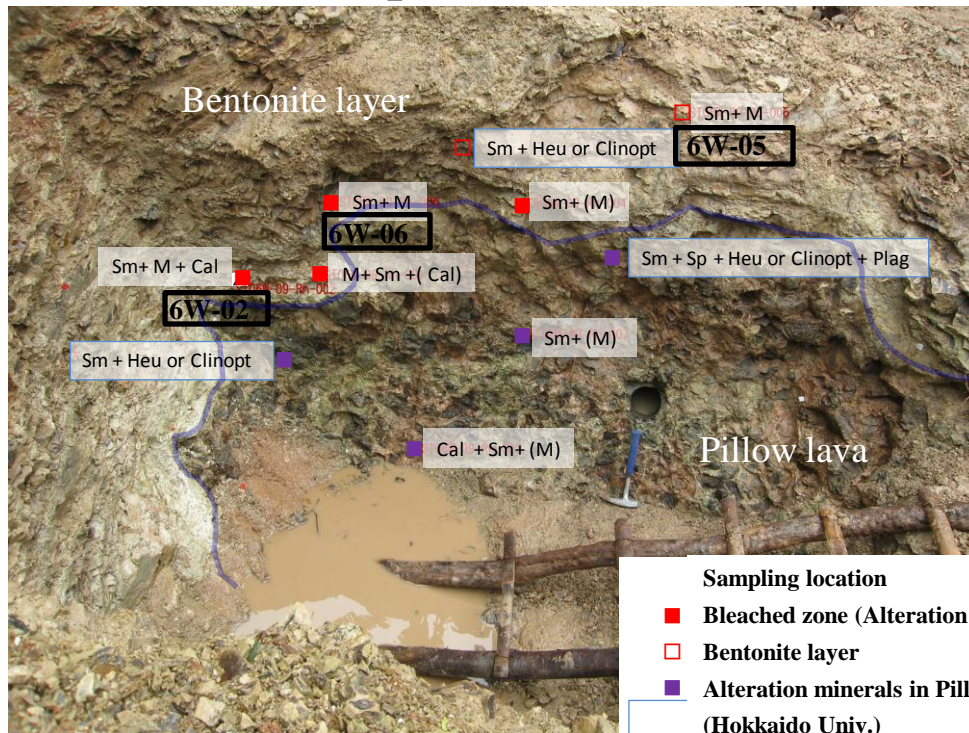
Sampling location



FESEM image of smectite

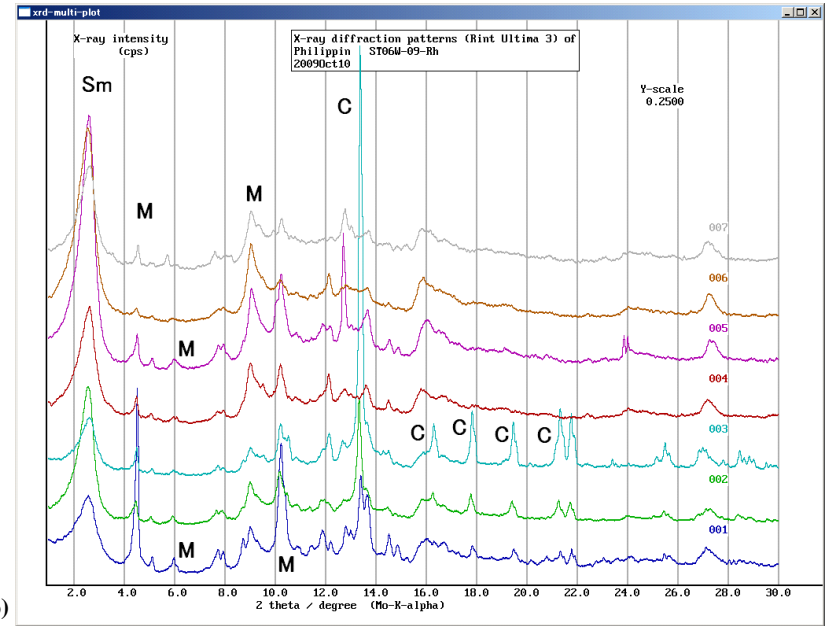
# 5. Mineralogical characterization of Bentonite

## 2. Mineral Composition of Trench 6 West Wall

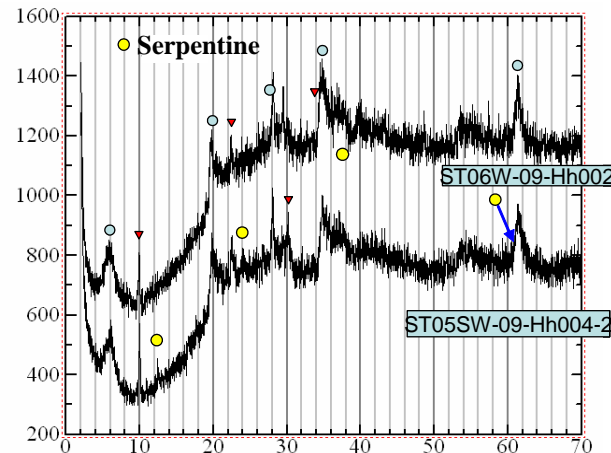


### Sampling location

- Bleached zone (Alteration Halo)
- Bentonite layer
- Alteration minerals in Pillow lava (Hokkaido Univ.)



Upper left: Sampling location  
 Upper right: XRD Results (Tokyo Institute of Tech.)  
 Lower right: XRD Results (Hokkaido Univ.)

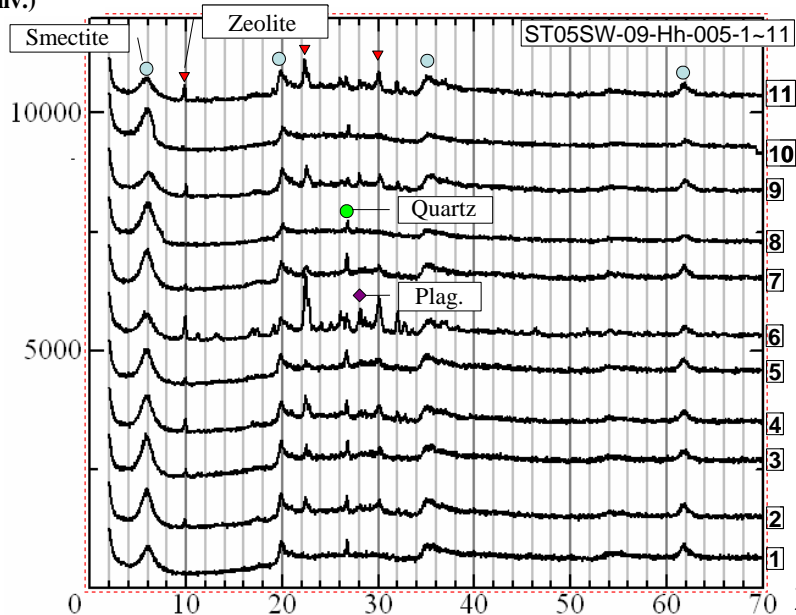
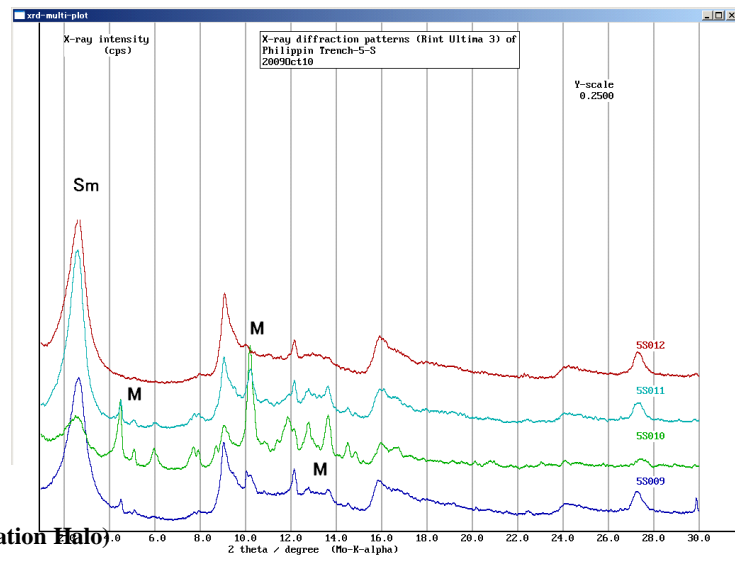
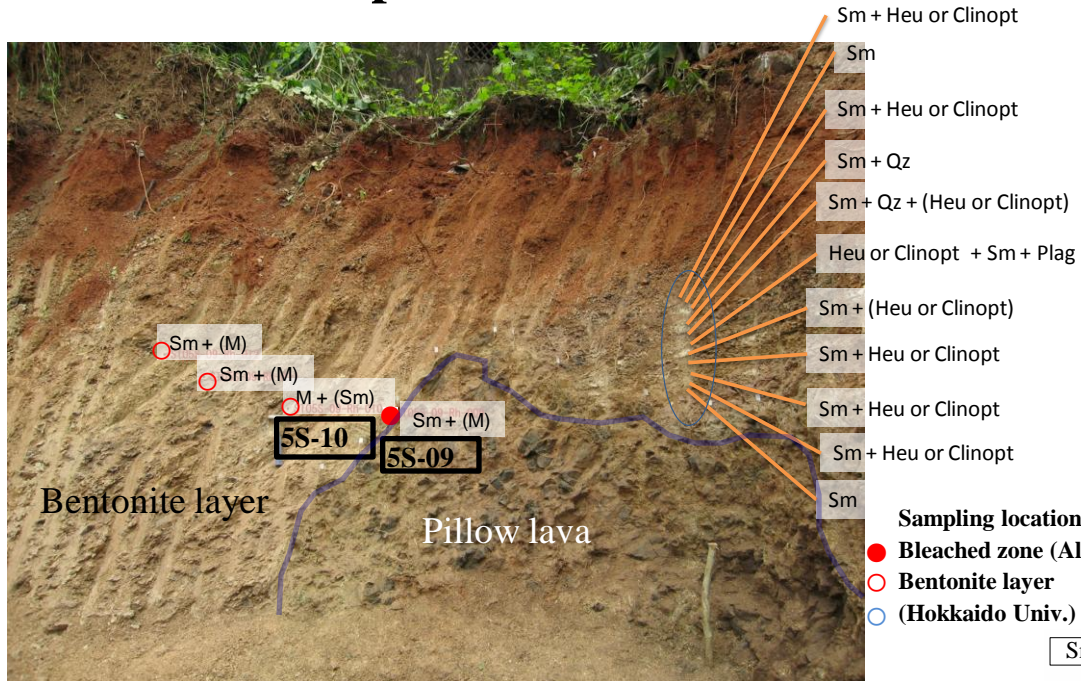


|                                 | Trench 5 (S·SW Wall)               | Trench 6 (W Wall)      |
|---------------------------------|------------------------------------|------------------------|
| Bentonite Layer                 | Sm+Zeo<br>Sm+Zeo+Qz+Plag<br>Sm+Zeo | Sm<br>Sm+Zeo<br>Sm+Cal |
| Bleached Zone (Alteration Halo) | Zeo+Sm<br>Sm+Zeo+Sp                | Zeo+Sm<br>Sm+Cal+Zeo   |
| Pillow Lava                     | Sm+Cal+Zeo+Sp                      | Sm+Zeo+Cal+Sp<br>+Plag |

Sm: Smectite, Zeo: Zeolite) Qz: Quartz, Cal: Calcite, Sp: Serpentine, Plag: Plagioclase

# 5. Mineralogical characterization of Bentonite

## 3. Mineral Composition of Trench 5 South Wall



Upper left: Sampling location

Upper right: XRD Results (Tokyo Institute of Tech.)

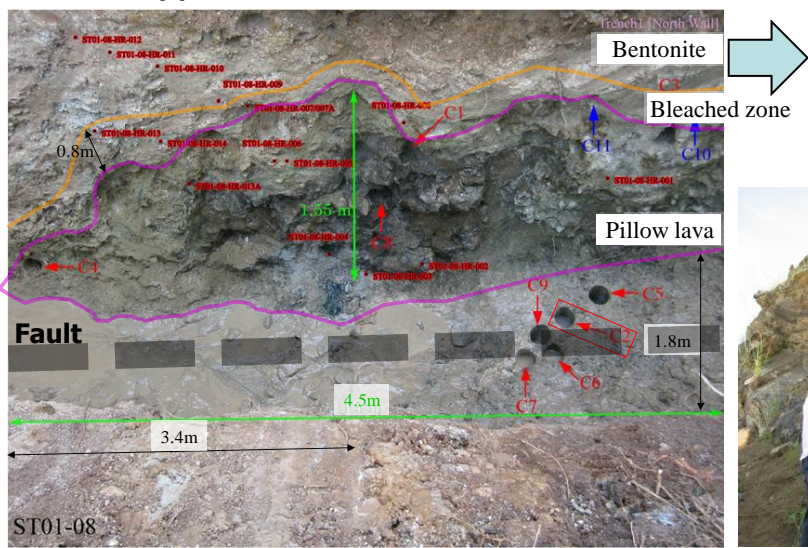
Lower right: XRD Results (Hokkaido Univ.)

|                                    | Trench 5<br>(S-SW Wall)            | Trench 6<br>(W Wall)   |
|------------------------------------|------------------------------------|------------------------|
| Bentonite Layer                    | Sm+Zeo<br>Sm+Zeo+Qz+Plag<br>Sm+Zeo | Sm<br>Sm+Zeo<br>Sm+Cal |
| Bleached Zone<br>(Alteration Halo) | Zeo+Sm<br>Sm+Zeo+Sp                | Zeo+Sm<br>Sm+Cal+Zeo   |
| Pillow Lava                        | Sm+Cal+Zeo+Sp                      | Sm+Zeo+Cal+Sp+Plag     |

Sm: Smectite, Zeo: Zeolite, Qz: Quartz, Cal: Calcite, Sp: Serpentine, Plag: Plagioclase

# 5. Conceptual Model on Pathway of the Past Hyperalkaline fluids (groundwater) formed as conduit filled with Serpentine, Mn-hydroxides, Carbonates and Ultramafic boulders (Fossil Type)

## Fossil Type of Past Interaction Zone

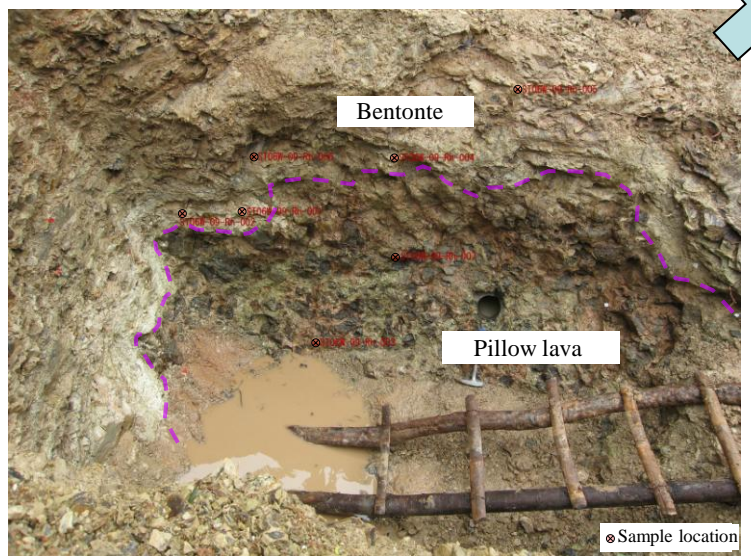
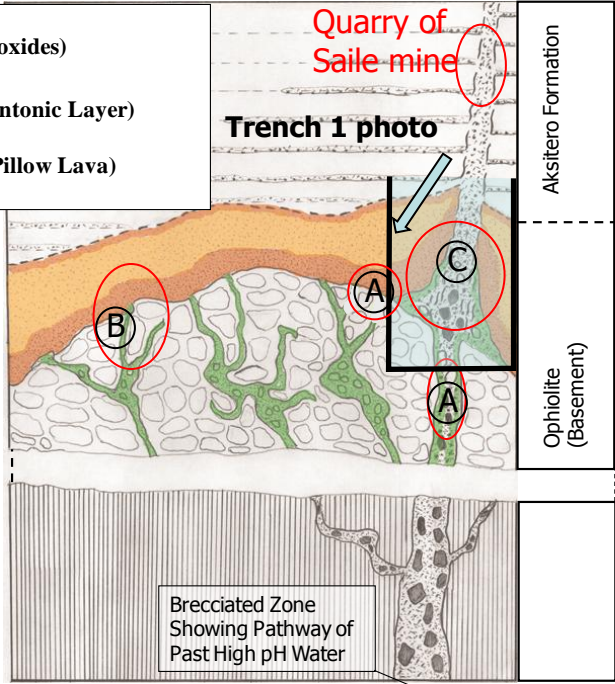


(A) Saile mine Trench 1 (2008)

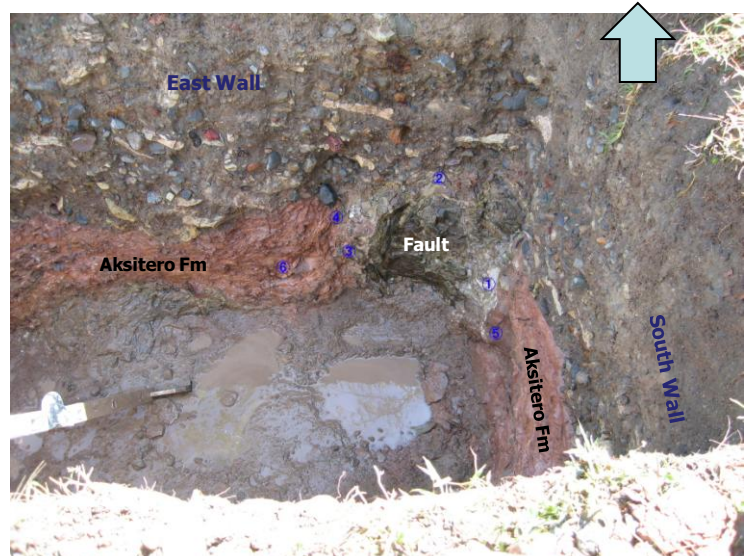
- Fracture & Fault
- Mn-Staining (Mn-Hydroxides)
- Bleached Zone (Strongly Altered Halo)
- Aksitero Formation (Bentonic Layer)
- Serpentinized Vein
- Ophiolite (Basement) (Pillow Lava)
- Gabbro (Dyke & Sheet)



Fracture system of quarry



(B) Saile mine Trench 6 (2009)

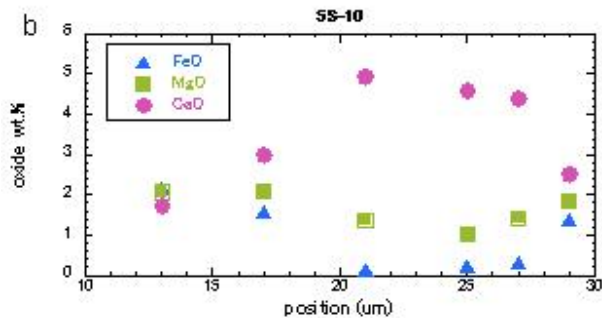
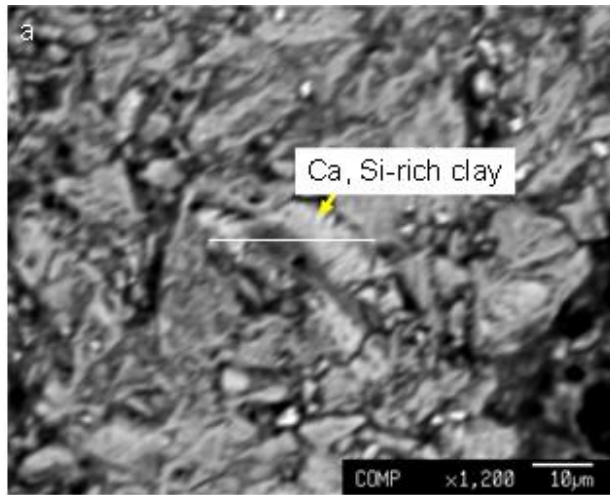


(C) Bigbiga Trench BBT01 (2010)

# 5. Mineralogical characterization of Bentonite

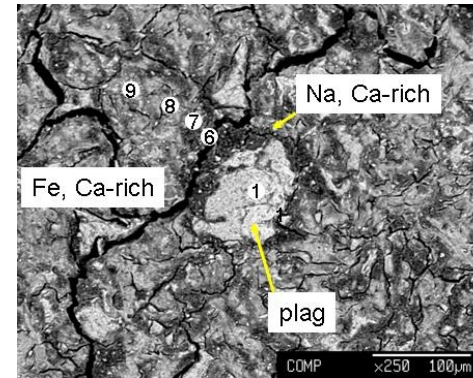
## Occurrence of smectite by FE-SEM analysis

(1) **Type 1**; Zoning (Ca·Si-rich (Core) and Fe·Mg rich (Rim)) of smectite(Parallel-growth)



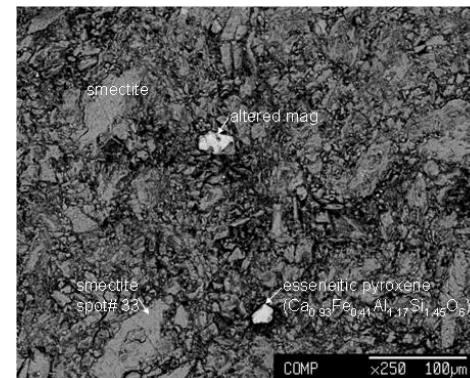
5S-10

(2) **Type 2**; Surrounding outside of Ca-rich core with Fe·Mg-rich smectite showing the shape of corona (Over-growth)



6W-05

(3) **Type 3**; Fe-rich smectite occurred as interstitial filling near Ca-rich smectite



5S-10

# 5. Chemical Composition of Bentonite

## Type 1

| spot#                          | 5S-09_35 | 5S-09_37 | 5S-09_alt40 | 5S-09_alt42 | 5S-09_alt43 |
|--------------------------------|----------|----------|-------------|-------------|-------------|
| Oxide wt.%                     |          |          |             |             |             |
| SiO <sub>2</sub>               | 61.79    | 55.91    | 52.07       | 42.97       | 41.87       |
| Al <sub>2</sub> O <sub>3</sub> | 16.51    | 13.95    | 16.15       | 12.12       | 10.65       |
| TiO <sub>2</sub>               | 0.22     | 0.30     | 0.46        | 0.29        | 0.78        |
| FeO                            | 11.71    | 11.73    | 13.32       | 16.02       | 15.94       |
| MnO                            | 0.22     | 0.08     | 1.83        | 10.53       | 14.32       |
| MgO                            | 3.58     | 3.24     | 3.27        | 2.51        | 2.31        |
| CaO                            | 1.65     | 1.64     | 1.64        | 1.20        | 0.93        |
| Na <sub>2</sub> O              | 0.06     | 0.16     | 0.10        | 0.04        | 0.23        |
| K <sub>2</sub> O               | 1.66     | 1.48     | 1.49        | 1.06        | 1.17        |
| Total                          | 97.39    | 88.48    | 90.33       | 86.74       | 88.20       |
| Cations based on O = 22 (apfu) |          |          |             |             |             |
| Si                             | 7.936    | 7.962    | 7.451       | 6.958       | 6.843       |
| Al(4)                          | 0.064    | 0.038    | 0.549       | 1.042       | 1.157       |
| sumT=8                         |          |          |             |             |             |
| Al(6)                          | 2.435    | 2.303    | 2.175       | 1.272       | 0.895       |
| Ti                             | 0.021    | 0.032    | 0.050       | 0.035       | 0.095       |
| Fe                             | 1.258    | 1.397    | 1.593       | 2.170       | 2.179       |
| Mn                             | 0.024    | 0.009    | 0.222       | 1.444       | 1.982       |
| Mg                             | 0.686    | 0.687    | 0.697       | 0.606       | 0.564       |
| sumM=4-6                       | 4.423    | 4.428    | 4.736       | 5.527       | 5.715       |
| XMg*                           | 0.353    | 0.330    | 0.304       | 0.218       | 0.206       |
| Ca                             | 0.227    | 0.250    | 0.251       | 0.208       | 0.162       |
| Na                             | 0.015    | 0.044    | 0.029       | 0.012       | 0.072       |
| K                              | 0.273    | 0.269    | 0.272       | 0.219       | 0.245       |
| sumIL                          | 0.514    | 0.564    | 0.551       | 0.439       | 0.479       |
| XCa**                          | 0.440    | 0.444    | 0.456       | 0.474       | 0.339       |

NOTE \*: XMg = Mg / (Mg + Fe)

\*\* : XCa = Ca / (Ca + Na + K)

## Type 2

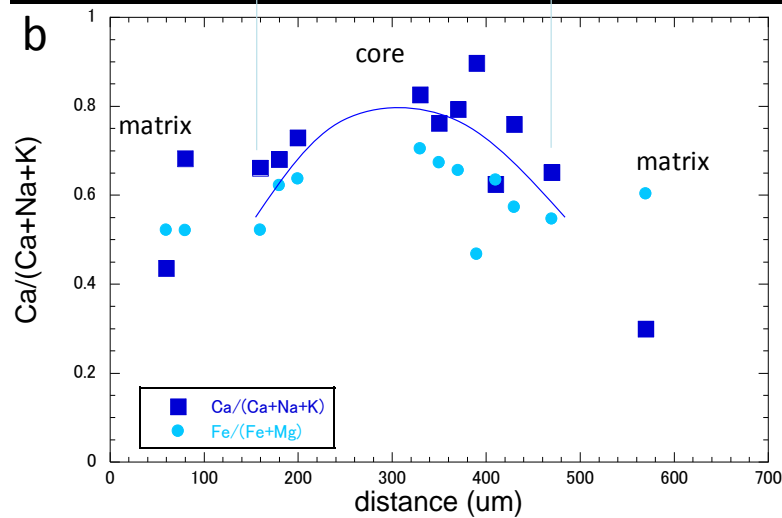
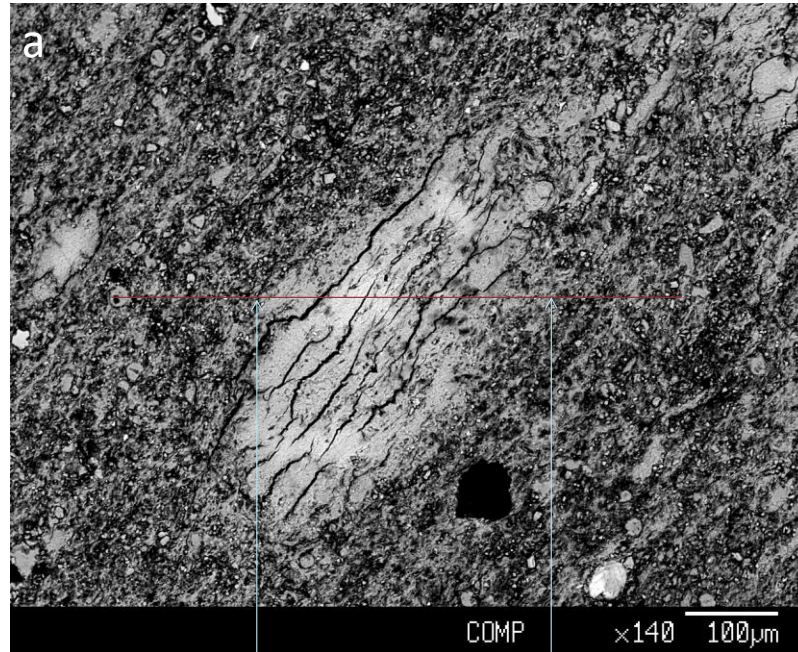
| spot#                          | 5S-10_001 | 5S-10_2 | 5S-10_4 | 5S-10_ca33 | 5S-10_35 |
|--------------------------------|-----------|---------|---------|------------|----------|
| Oxide wt.%                     |           |         |         |            |          |
| SiO <sub>2</sub>               | 55.71     | 56.23   | 52.04   | 60.26      | 56.87    |
| Al <sub>2</sub> O <sub>3</sub> | 22.22     | 22.20   | 22.20   | 27.52      | 24.76    |
| TiO <sub>2</sub>               | 0.41      | 0.59    | 0.34    | 0.25       | 0.80     |
| FeO                            | 4.24      | 5.18    | 4.62    | 3.65       | 5.74     |
| MnO                            | 0.09      | 0.03    | 0.00    | 0.00       | 0.10     |
| MgO                            | 3.58      | 3.66    | 2.69    | 2.21       | 2.62     |
| CaO                            | 1.62      | 1.66    | 1.33    | 0.94       | 1.64     |
| Na <sub>2</sub> O              | 0.07      | 0.10    | 0.11    | 0.06       | 0.09     |
| K <sub>2</sub> O               | 0.69      | 1.08    | 1.24    | 1.09       | 0.44     |
| Total                          | 88.63     | 90.72   | 84.56   | 95.98      | 93.06    |
| Cations based on O = 22 (apfu) |           |         |         |            |          |
| Si                             | 7.542     | 7.499   | 7.433   | 7.462      | 7.375    |
| Al(4)                          | 0.458     | 0.501   | 0.567   | 0.538      | 0.625    |
| sumT=8                         |           |         |         |            |          |
| Al(6)                          | 3.087     | 2.988   | 3.170   | 3.479      | 3.160    |
| Ti                             | 0.041     | 0.059   | 0.036   | 0.023      | 0.078    |
| Fe                             | 0.480     | 0.577   | 0.552   | 0.378      | 0.623    |
| Mn                             | 0.011     | 0.004   | 0.000   | 0.000      | 0.011    |
| Mg                             | 0.723     | 0.728   | 0.573   | 0.408      | 0.507    |
| sumM=4-6                       | 4.341     | 4.356   | 4.331   | 4.288      | 4.379    |
| XMg*                           | 0.601     | 0.558   | 0.509   | 0.519      | 0.449    |
| Ca                             | 0.235     | 0.237   | 0.204   | 0.125      | 0.227    |
| Na                             | 0.019     | 0.025   | 0.030   | 0.015      | 0.023    |
| K                              | 0.119     | 0.183   | 0.226   | 0.172      | 0.072    |
| sumIL                          | 0.373     | 0.446   | 0.460   | 0.312      | 0.322    |
| XCa**                          | 0.630     | 0.532   | 0.444   | 0.401      | 0.706    |

NOTE \*: XMg = Mg / (Mg + Fe)

\*\* : XCa = Ca / (Ca + Na + K)

# 5. Zonig profile

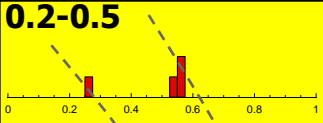
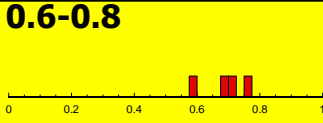
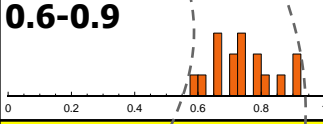
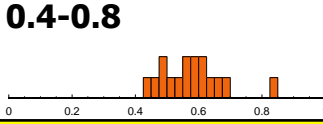
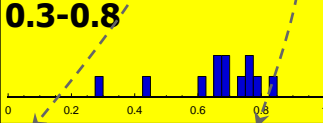
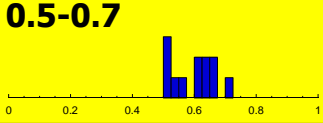
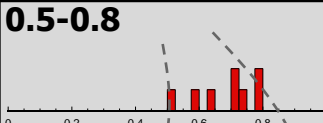
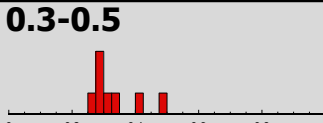
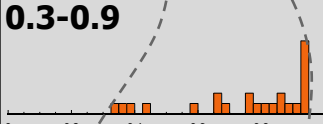
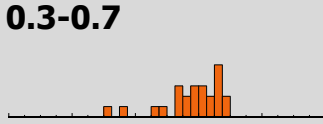
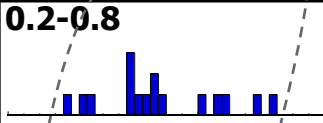
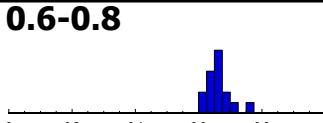
## Zoning profile (Saile mine Trench 1(Bleaced))



ST-012

# 5. Composition change of bentonite

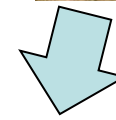
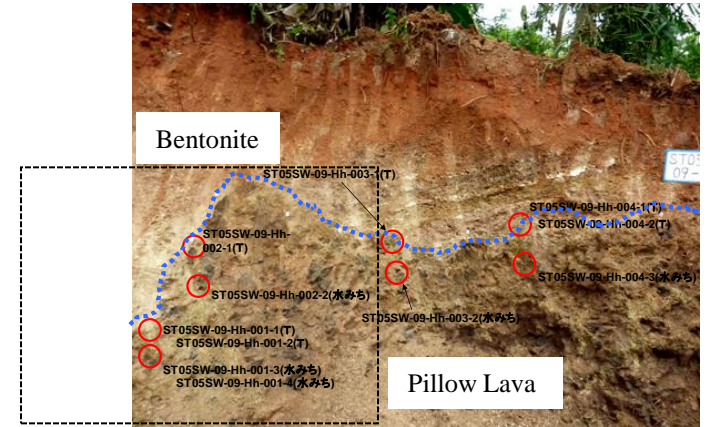
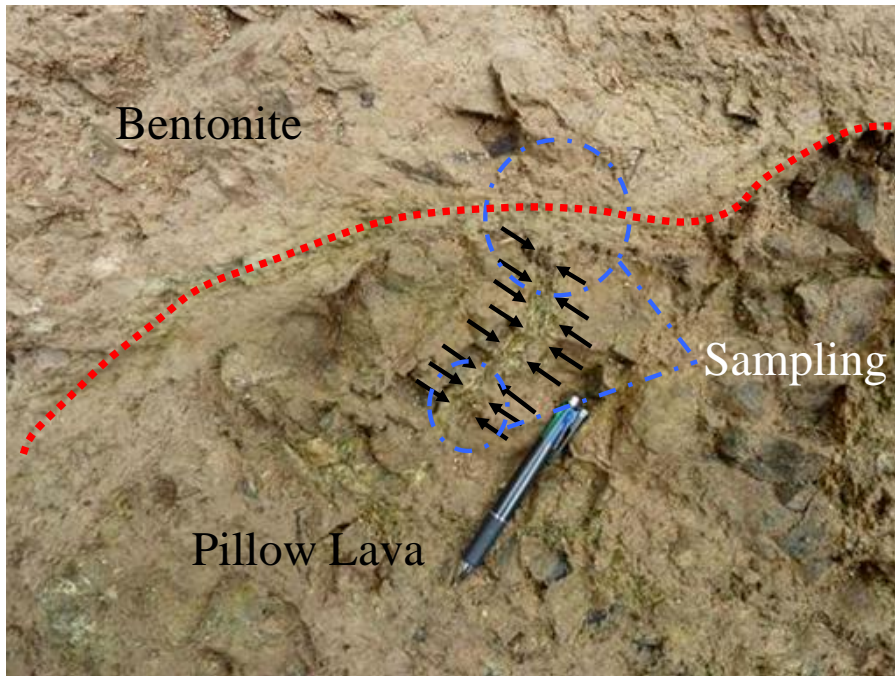
## Summary of FE-SEM analysis

| Sample ID  | Alteration minerals | Relict minerals                     | Zeolite composition                  | Smectite XCa-composition  | Smectite XFe-composition  |
|--|---------------------|-------------------------------------|--------------------------------------|---|---|
| Trench-1 (Saile Mine) <span style="float: right;">※ Yellow: Bleached Zone</span> |                     |                                     |                                      |   |   |
| ST-001   | Zeolite >> Smectite | Plag, Cpx, Opx, Qz                  | Mordenite - Si-rich stellerite       | <b>0.2-0.5</b><br>   | <b>0.6-0.8</b><br>   |
| ST-003   | Smectite >> Zeolite | Volcanic clast                      |                                      |   |   |
| ST-006   | Smectite >> Zeolite | Plag, Qz                            |                                      |   |   |
| ST-007   | Smectite > Zeolite  | Plag, Cpx, Qz                       | Yugawaralite                         | <b>0.6-0.9</b><br>   | <b>0.4-0.8</b><br>   |
| ST-008   | Smectite >> Zeolite | Plag, Cpx, Qz                       |                                      |   |   |
| ST-012   | Smectite > Zeolite  | Plag, Cpx, Qz                       | Clinoptilolite - Si-rich stellerite  | <b>0.3-0.8</b><br>   | <b>0.5-0.7</b><br>   |
| Bigbiga Trench-1 <span style="float: right;">※※ Gray: Fractured Zone</span>      |                     |                                     |                                      |   |   |
| BB-001   | Zeolite >> Smectite | Plag, Cpx, Qz                       | Heulandite – Yugawaralite-Stellerite | <b>0.5-0.8</b><br>  | <b>0.3-0.5</b><br>  |
| BB-002   | Zeolite >> Smectite | Plag, Cpx, Qz, Calc, Ba-phase       | Yougawaralite - Stellerite           | <b>0.3-0.9</b><br> | <b>0.3-0.7</b><br> |
| BB-004   | Smectite > Zeolite  | Plag, Cpx, Qz, Calc (Mn-rich), Apa, | Mordenite - Stellerite               | <b>0.2-0.8</b><br> | <b>0.6-0.8</b><br> |
| BB-006   | Smectite > Zeolite  | Plag, Cpx, Qz, Calc, Apa, Mn-oxide  | Stellerite                           |   |   |

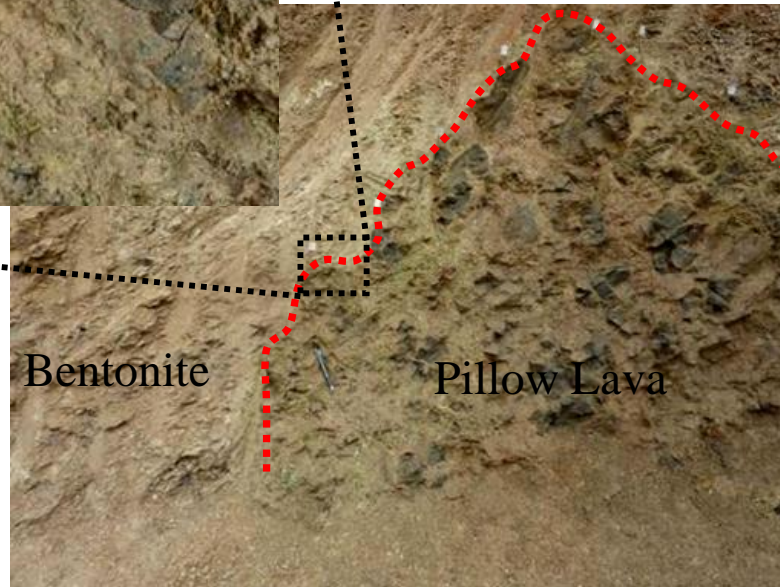
# 6. Interaction between smectite and hyperalkaline groundwater

## Lithology of Alteration Zone between Pillow lava and Bentonite layer

### (1) Sampling (Trench 5) (Hokkaido Univ.)



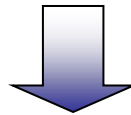
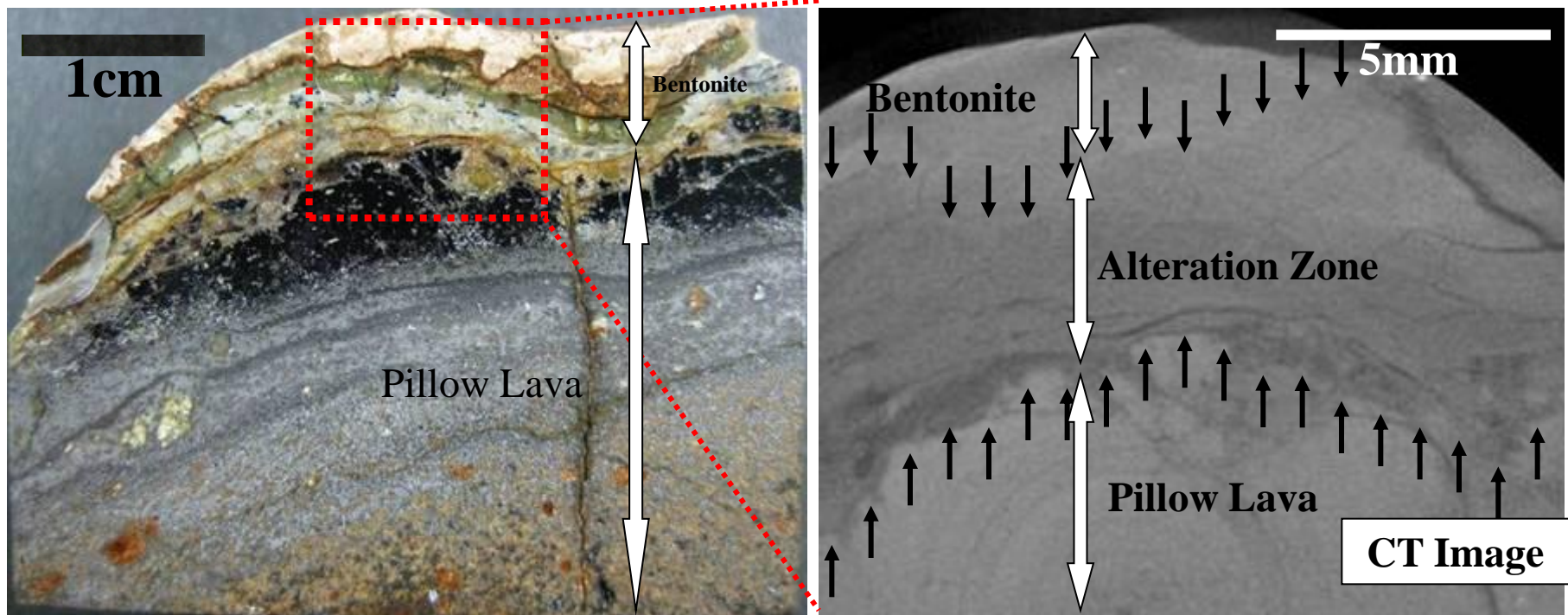
Trench 5 SW Wall



# 6. Interaction between smectite and hyperalkaline groundwater

## Detailed Structure of Alteration Zone

### (2) Alteration of bentonite in mm scale (Hokkaido Univ.)

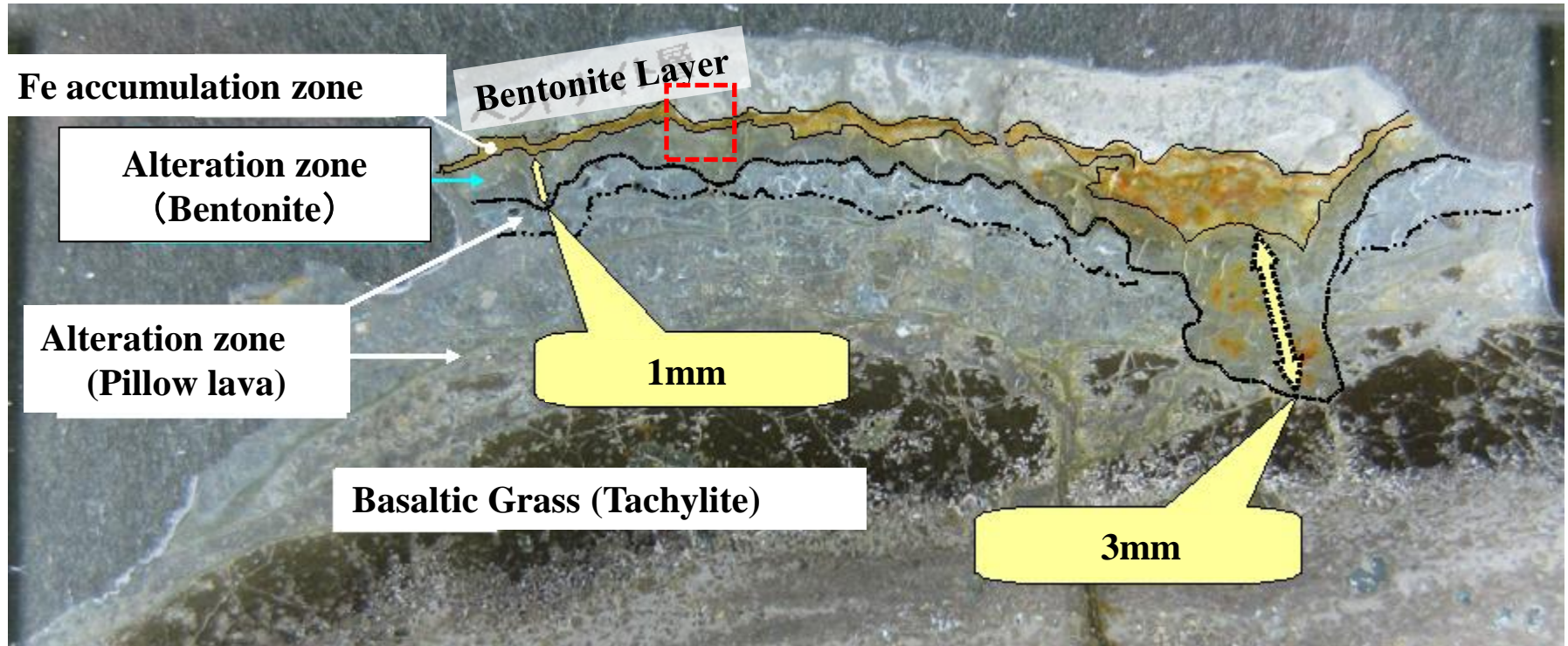


The density of alteration zone differs from bentonite and pillow lava.

# 6. Interaction between smectite and hyperalkaline groundwater

## Detailed Structure of Alteration Zone

### (3) Alteration of bentonite in mm scale (Hokkaido Univ.)

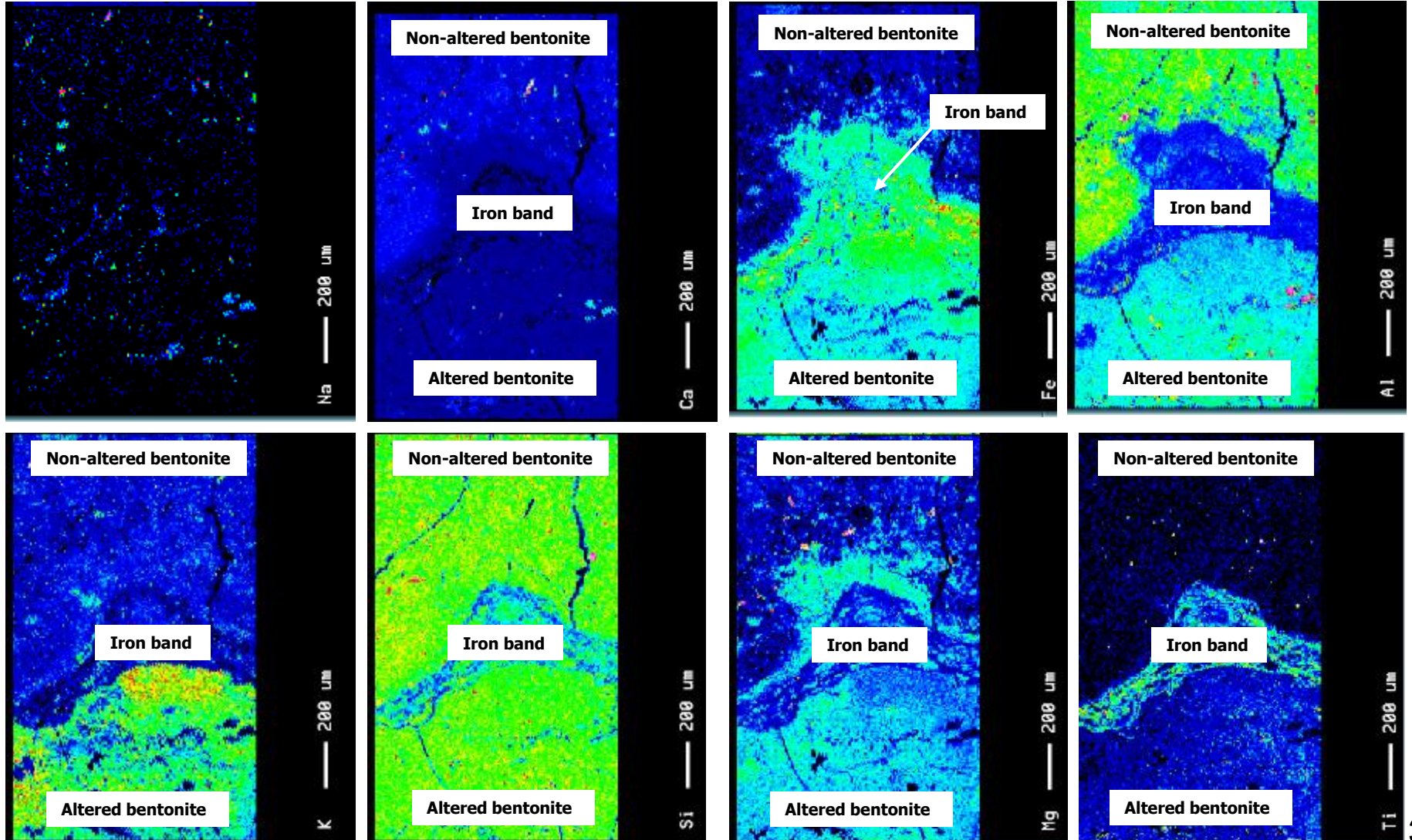


- Alteration zone between bentonite and pillow lava has been identified near the contact.
- Fe accumulation zone can be seen at the front of alteration zone.
- Width of alteration zone of bentonite is 1 - 3mm.

# 6. Interaction between smectite and hyperalkaline groundwater

## EPMA Mapping

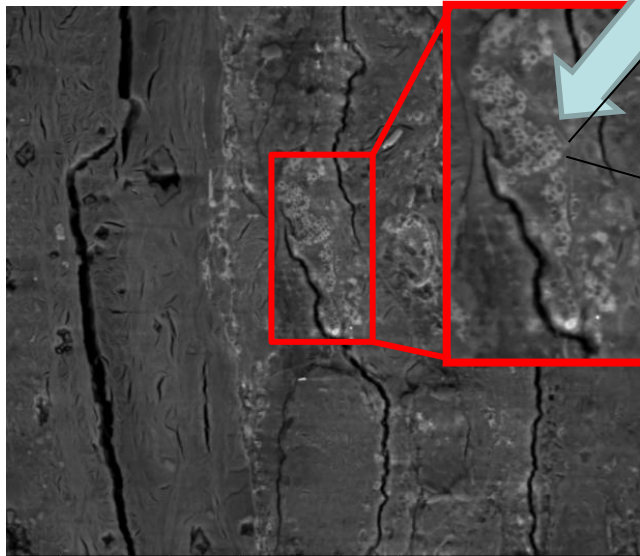
(Domain encircled by the red dotted line of the pre-slide )



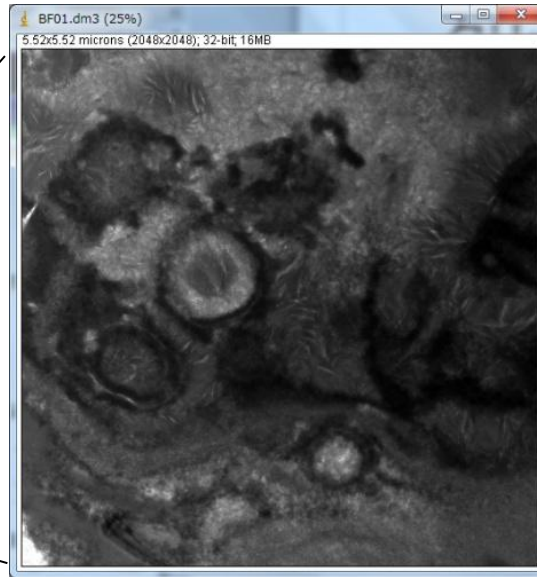
# 6. Mineralogical characterization of iron accumulation band in altered bentonite

SEM observation of iron band

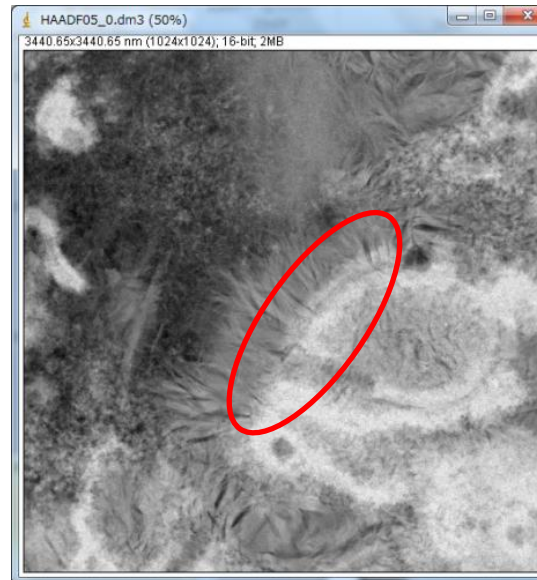
Grain structure



×4



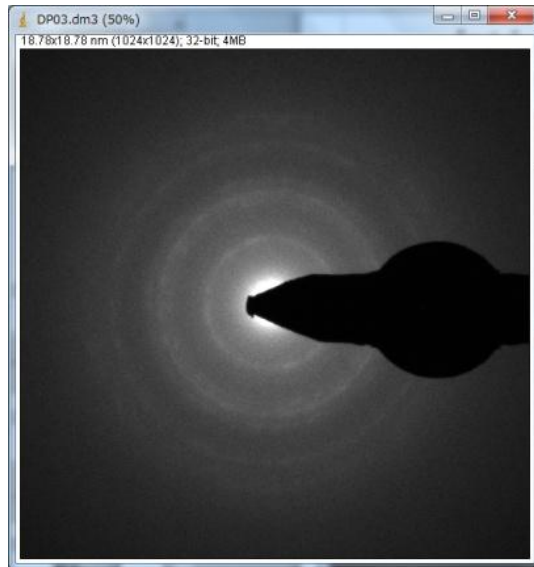
Grain structure



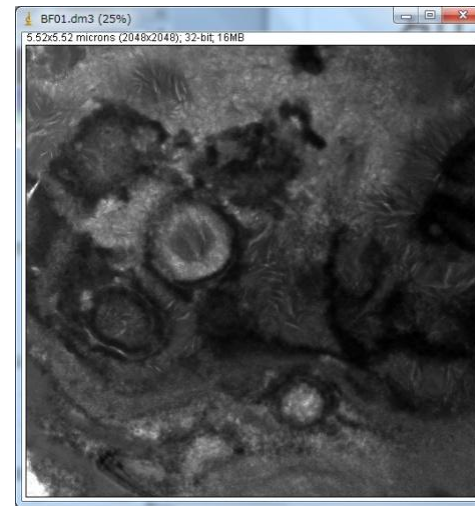
Feather structure

# 6. Mineralogical characterization of iron accumulation band in altered bentonite

Electron diffraction analysis



Electron diffraction pattern of Grain point



Bright part : Fe and Ti rich

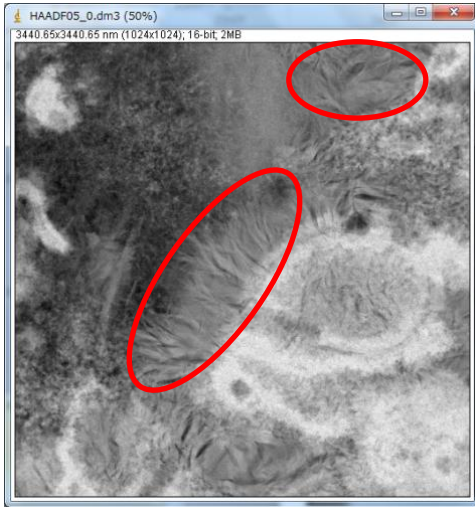
|        |       | nm    |
|--------|-------|-------|
| 4.735  | 2.367 | 0.42  |
| 7.58   | 3.79  | 0.26  |
| 8.277  | 4.139 | 0.24  |
| 11.636 | 5.818 | 0.171 |



Correspondence with XRD pattern of **Goethite**

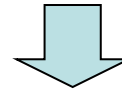
# 6. Nontronitization of Ca-Smectite

## Electron diffraction analysis

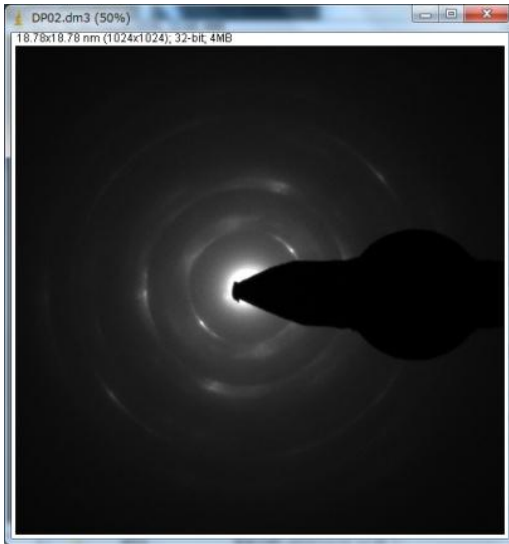


|        |       | nm    |
|--------|-------|-------|
| 4.442  | 2.221 | 0.45  |
| 6.13   | 3.06  | 0.33  |
| 7.819  | 3.909 | 0.26  |
| 8.993  | 4.496 | 0.22  |
| 13.343 | 6.671 | 0.149 |

Correspondence with XRD pattern of smectite



**Nontronite** from point analysis of Fe accumulation band by EPMA

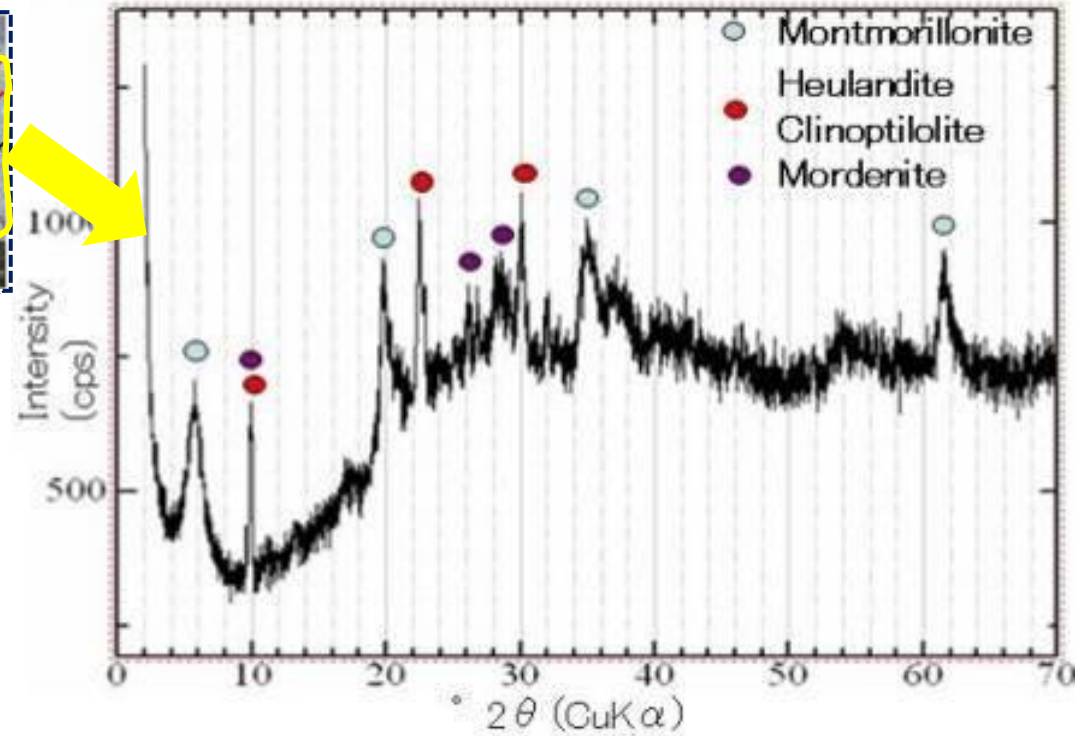
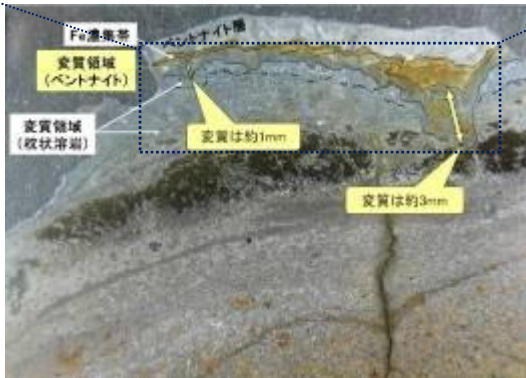


Electron diffraction pattern of Feather structure

| Element | 8      | 9      |
|---------|--------|--------|
| Na      | 0.0082 | -      |
| Ca      | 0.4699 | 0.4617 |
| Fe      | 2.5484 | 2.319  |
| Mg      | 1.1065 | 1.3742 |
| K       | 0.0584 | 0.1315 |
| Mn      | 0.1498 | 0.1072 |
| Al      | 1.0322 | 1.1721 |
| Ti      | 0.0084 | 0.0149 |
| Si      | 9.0636 | 8.9421 |

# 6. Interaction between smectite and hyperalkaline groundwater

XRD(Yellow line)



## 6. Interaction between smectite and hyperalkaline groundwater

Comparison with the precedence research on alteration bentonite

|                                | Mo    | Na-Cp | Ca-Cp | K-Cp  | Ca-H  | K-H   | Wt% |
|--------------------------------|-------|-------|-------|-------|-------|-------|-----|
| SiO <sub>2</sub>               | 61.30 | 69.52 | 66.78 | 66.25 | 61.28 | 61.75 |     |
| Al <sub>2</sub> O <sub>3</sub> | 14.25 | 13.18 | 13.27 | 12.70 | 14.95 | 16.46 |     |
| FeO                            | 2.90  | 0.00  | 0.17  | 0.11  | 0.27  | 0.09  |     |
| MgO                            | 0.80  | 0.00  | 0.27  | 0.19  | 0.41  | 0.02  |     |
| CaO                            | 3.64  | 0.13  | 2.81  | 0.14  | 3.93  | 0.19  |     |
| Na <sub>2</sub> O              | 1.54  | 7.07  | 2.81  | 0.75  | 1.84  | 0.14  |     |
| K <sub>2</sub> O               | 2.80  | 0.61  | 1.16  | 9.94  | 1.00  | 10.50 |     |

|                                | Phillipine | Wt% |
|--------------------------------|------------|-----|
| SiO <sub>2</sub>               | 42.98      |     |
| Al <sub>2</sub> O <sub>3</sub> | 9.72       |     |
| FeO                            | 16.10      |     |
| MgO                            | 3.90       |     |
| CaO                            | 2.67       |     |
| Na <sub>2</sub> O              | 0.05       |     |
| K <sub>2</sub> O               | 2.14       |     |

Mo: Mordenite

(Ogiwara 1996)

Na-Cp: Na-Clinoptilolite

Ca-H: Ca-Heulandite

Ca-Cp: Ca-Clinoptilolite

K-H: K-Heulandite

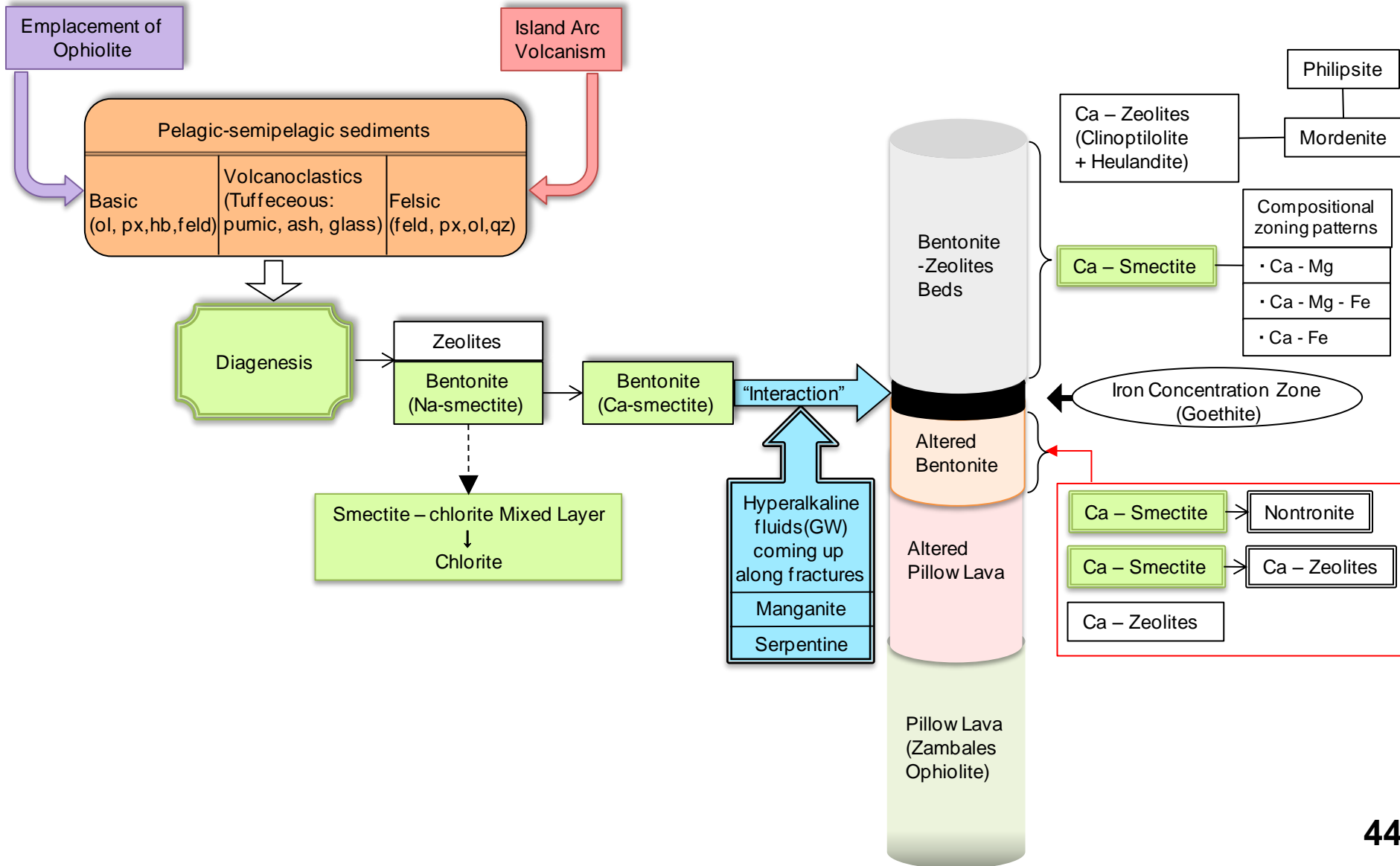
K-Cp: K-Clinoptilolite

- XRD suggests that alteration bentonite is likely to Clinoptilolite or Heulandite.
- Boles (1972) defined zeolites of Si/Al>4 and Si/Al<4 are Clinoptilolite and Heulandite, respectively.
- It was reported that Clinoptilolite was generally formed result from reaction between Heulandite and Ca rich water.



But, the kind of zeolite cannot have been identified yet only with such evidence.

# 7. Alteration Process of Bentonite in the Philippines NA site



# 8. Summary and Future plan

## Summary

- Some clear evidences of contact with a bentonite and hyperalkaline groundwater in the past were found out in the Philippines. (Fossil Type of Natural Analogue)
- XRD does not show the alkaline alteration of bentonite in the bleached zone. That is the bentonite maintains enough smectite.
- The characteristic occurrences of the smectite observed by FESEM are characterized by 3 types.
- From the basalt glass (Tachylite) where is a outermost part of pillow lava,  $Fe^{2+}$  oxideized and Si, Al, Na and Ca were selectively leached, cosequently the alteration zone of about 5mm was formed.
- The alteration zone of bentonite consists of smectite and Ca-Zeolite (Heulandite, Clinoptilolite, Mordenite) .
- The Fe-accumulated band consisted of mainly Goethite is located between bleached zone and unaltered bentonite bed. The width of this zone is about 1~3 mm. It is only observed notronite to occur adjacent to this zone.

## Future plan

- The presence of phases formed by typical hyperalkaline alteration of smectites was not found, but this has to be investigated further.
- It is necessary to obtain chronological data to make clear the time-scales of Bentonite Reaction, and to establish scenario for Bentonite alteration in high pH conditions.
- This program should be focused on studying not only “Fossil site (Saile mine)” but also “Active site (Bigbiga)” with ongoing reaction between Bentontie and Hyperalkaline groundwater.